

Limiting Reagent and Percentage Yield

- 1 Dichlorine monoxide is produced by passing chlorine gas over heated mercury (II) oxide according to the following unbalanced reaction:



Balance the reaction. Suppose the quantity of the reactants is sufficient to produce 0.86 g of Cl_2O , but only 0.71 g is obtained. What is the percentage yield?

- 2 Consider the following reaction:



The theoretical yield of beryllium chloride was 10.7 g. Suppose the reaction actually yields 4.5 g. What is the percentage yield?

- 3 Nitrogen and hydrogen at high temperatures are converted to ammonia using the following reaction:



When 400 g of H_2 are added to an excess amount of N_2 , 104 g of NH_3 are formed. Calculate the percentage yield.

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General, Organic, and Biological Chemistry Michael P. Garoutte, 2014-02-24 Classroom activities to support a General Organic and Biological Chemistry text Students can follow a guided inquiry approach as they learn chemistry in the classroom General Organic and Biological Chemistry A Guided Inquiry serves as an accompaniment to a GOB Chemistry text It can suit the one or two semester course This supplemental text supports Process Oriented Guided Inquiry Learning POGIL which is a student focused group learning philosophy of instruction The materials offer ways to promote a student centered science classroom with activities The goal is for students to gain a greater understanding of chemistry through exploration

Chemical News and Journal of Industrial Science, 1912 *Engineering; an Illustrated Weekly Journal*, 1917

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Engineering, 1917 **Engineering Index Annual**, 1948 **A Supplement to the Oxford English Dictionary:**

O-Scz R. W. Burchfield, 1972 Supplement to the Oxford dictionary of the English language comprising new words and senses of the period from 1884 to the present day replaces the earlier 1933 supplement **The Oxford English Dictionary**, 1989

Journal American Chemical Society, 2001 **Agrindex**, 1995

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