

**PART - I**

15 X 1 = 15

**I Answer all the questions.**

1. Which of the following is / are true with respect to Carbon - 12  
a) relative atomic mass is 12u      b) oxidation number of Carbon is +4 in all its compounds  
c) mole of Carbon - 12 contain  $6.022 \times 10^{22}$  carbon atoms      d) all of these
2. How many electrons in an atom with atomic number 105 can have  $(n + l) = 8$   
a) 30      b) 17      c) 15      d) unpredictable
3. Among the following which is isoelectronic with  $O^{2-}$  ion,  
a)  $Cl^-$       b)  $F^-$       c)  $Ca^{2+}$       d)  $N^{2-}$
4. The cause of permanent hardness of water is due to  
a)  $Ca(HCO_3)_2$       b)  $Mg(HCO_3)_2$       c)  $CaCl_2$       d)  $MgCO_3$
5. Sodium is stored in  
a) Alcohol      b) Water      c) Kerosene      d) None of these.
6. If temperature and volume of an ideal gas is increased to twice its values, the initial pressure P becomes  
a) 4p      b) 2p      c) p      d) 3p
7. Heat of combustion is always  
a) Positive      b) negative      c) zero      d) either positive or negative
8. In the reaction,  $Fe(OH)_3(s) \rightleftharpoons Fe^{3+}(aq) + 3OH^-(aq)$  if the concentration of  $OH^-$  ions is decreased by  $1/4$  times, then the equilibrium concentration of  $Fe^{3+}$  will  
a) not changed      b) also decreased by  $1/4$  times  
c) increase by 4 times      d) increase by 64 times
9. According to Raoult's law, the relative lowering of vapour pressure for a solution is equal to  
a) mole fraction of solvent      b) mole fraction of solute  
c) number of moles of solute      d) number of moles of solvent
10. Number of sigma ( $\sigma$ ) and pi ( $\pi$ ) bonds in acetylene is .....  
a)  $2\sigma$  &  $2\pi$       b)  $3\sigma$  &  $3\pi$       c)  $2\sigma$  &  $3\pi$       d)  $3\sigma$  &  $2\pi$
11. Sodium nitroprusside reacts with sulphide ion to give a purple colour due to the formation of  
a)  $[Fe(CN)_5 NO]^{3-}$       b)  $[Fe(NO)_5 CN]^+$       c)  $[Fe(CN)_5 NOS]^{4-}$       d)  $[Fe(CN)_5 NOS]^{3-}$
12. What is the hybridisation state of benzyl carbonium ion?  
a)  $sp^2$       b)  $sp^3$       c)  $sp^2$       d)  $sp^2$
13. .... is used for welding and cutting metals.  
a) Ethylene      b) Propylene      c) Poly ethylene      d) Oxy - acetylene
14. What should be the correct IUPAC name of diethyl chloromethane?  
a) 3 - chloro pentane      b) 1 - chloropentane  
c) 1 - chloro - 1, 1, diethyl methane      d) 1 - chloro - 1 - ethylethane
15. The pH of normal rain water is .....  
a) 6.5      b) 7.5      c) 5.6      d) 4.6

**PART - II**

6 x 2 = 12

**II Answer any six. (q.No. 24 is compulsory).**

16. Define modern periodic law.
17. What is water - gas shift reaction?

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**Clifford Lane**



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