

Progressive Discipline: What It Is, What It Isn't and How to More Successfully Use It with Your Employees



Presenter: Andrew Sanderbeck

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web oxidation reduction redox reactions oxidation $\text{Cu} \rightarrow \text{Cu}^{2+} + 2\text{e}^{-}$ reduction $\text{MnO}_4^{-} + 4\text{H}^{+} + 3\text{e}^{-} \rightarrow \text{MnO}_2 + 2\text{H}_2\text{O}$
 l when the half reactions listed above are combined into a balanced chemical equation the ratio of the coefficients $\text{Cu} : \text{MnO}_4^{-}$ is 4 : 1

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 3 aq $\text{Br}_2 + \text{Br}^{-} \rightarrow \text{Br}_2 + \text{Br}^{-}$ previous topic next topic

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