

g) A solution has pH 4.2596. Find out hydrogen ion concentration.

h) Define the terms

i) gangue

ii) ore

i) State two purposes of calcination.

j) Write composition of duralumin.

k) What is tack property of rubber?

l) Give two uses of thermocole based on its two different properties.

2. Attempt any **FOUR** of the following:

16

a) Distinguish between atomic number and atomic mass number.

b) Write the electronic configuration of the following (in s, p, d, f orbitals)  $2\text{He}^4$ ,  $6\text{C}^{12}$ ,  $12\text{Mg}^{24}$ ,  $19\text{K}^{39}$ .

c) Define covalency. Describe the formation of Nitrogen Molecule.

d) Define degree of ionization. Describe the factors affecting degree of ionization.

e) Describe electro-refining process with sketch.

f) A current of 2 amperes was passed through a solution of copper sulphate for 16 minutes and 5 seconds. Calculate the amount of copper deposited on cathode.

(At.wt. of Cu = 63.5)

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**E Durkheim**



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