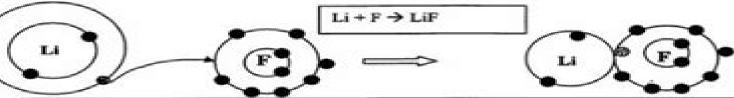
# Tonic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms Li Valence electrons

Electron transfer from/to each atom

Ions formed in the product

Li I

I from Li

F - lence Electron transfer lons form

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) Li +ci => Lici	,Li	1	1 to C1	Li+
	CI	フ	I from Li	CI-
2) C3 + 20 = C3 O	Ca	2	2+00	Ca <sup>2+</sup>
	0	6	2 from Ca	O <sup>2-</sup>
3) Be + F => BeF <sub>2</sub>	Be	2	1 to each F	Be2+
	F	7	Each 1 gcts 1 from Be	FT
4) Mg + S -> MgS	Mg	2	2 to S	Mg 2+
	S	6	2 from Mg	Sz-
5) K + E S KE	K	1	1 to F	K+2
	F	7	1 from K	F-1

# **Ionic Bonds Section 7 2 Worksheet Answers**

**Thomas Griffiths** 

#### **Ionic Bonds Section 7 2 Worksheet Answers:**

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