#### Introduction to Chemical Engineering

#### Final Exam

Name		
	•	

- T or F
   Enthalpy is comprised of enthalpy and the work an object must do to
  move
  the surroundings out of its way.
- T or F The velocity profile for laminar flow of a Newtonian fluid in a circular pipe
   appears as a parabola.
- T or F Fourier's Law of Diffusion relates diffusion to a mass gradient. (i.e. from a low concentration to a low concentration)
- T or F Internal Energy is comprised of enthalpy and the work an object must do
  to
  move the surroundings out of its way.
- 5. T or F The velocity profile for laminar flow of a Newtonian fluid in a circular pipe indicates that the fluid molecules in the center of the pipe are moving faster relative to the molecules that are closer to the pipe wall.
- T or F Fick's Law of Diffusion relates diffusion to a mass gradient. (i.e. from a high concentration to a low concentration)
- T or F To determine the activation energy of a reaction, experiments must be performed for at least three different temperatures.
- T or F A fluidized catalytic cracking reactor is used to remove sulfur from organic sulfur molecules.
- 9. Mass and Energy are:
  - a. conserved only for well-defined systems that are insulated such that there is no heat loss with surroundings.
  - conserved only for open systems (i.e. heat and mass flow across a system boundary)
  - balanced between flows that go into and out of a system such that each is conserved and are independent of whether it is an open system or a closed system.

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