

18. Lattice enthalpies at 298 K (experimental values)

The lattice enthalpy values ($\Delta H_{\text{lattice}}^\circ$) given relate to the endothermic process $M_aX_b(s) \rightarrow aM^{b+}(g) + bX^{a-}(g)$ in which the gaseous ions of a crystal are separated to an infinite distance from each other.

Experimental values

The data in these tables are experimental values obtained by means of a suitable Born-Haber cycle.

Alkali metal halides	$\Delta H_{\text{lattice}}^\circ$ (kJ mol ⁻¹)			
	F	Cl	Br	I
Li	1049	864	820	764
Na	930	790	754	705
K	829	720	691	650
Rb	795	695	668	632
Cs	759	670	647	613

Other substances	$\Delta H_{\text{lattice}}^\circ$ (kJ mol ⁻¹)
CaF ₂	2651
BeCl ₂	3033
MgCl ₂	2540
CaCl ₂	2271
SrCl ₂	2170
BaCl ₂	2069
MgO	3791
CaO	3401

Other substances	$\Delta H_{\text{lattice}}^\circ$ (kJ mol ⁻¹)
SrO	3223
BaO	3054
CuCl ₂	2824
AgF	974
AgCl	918
AgBr	905
AgI	892

Ib Chemistry Data Let

Albert A Gayle



Ib Chemistry Data Let:

Ib Chemistry Data Let Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Ib Chemistry Data Let**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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