

► **Alloys (page 41)**

9. A mixture of two or more elements that has the properties of a metal is called a(n) alloy.
10. Bronze is a mixture of what two metals? copper and tin

► **Metals in the Periodic Table (pages 41–44)**

11. How do the properties of each family of metals change as you move across the table? Metals tend to become less reactive as you move from left to right across the periodic table.
12. Circle the letter of each sentence that is true about alkali metals.
- ☒ a. They are never found as elements but only in compounds.
  - ☒ b. As elements, they are soft and shiny.
  - ☐ c. They are often found as pure elements in sea water.
  - ☒ d. They are extremely reactive.
13. What are the two most important alkali metals?  
sodium and potassium
14. Circle the letter of each sentence that is true about alkaline earth metals.
- ☒ a. They are good conductors of electricity.
  - ☒ b. They are never found uncombined in nature.
  - ☒ c. They are bright white and fairly hard.
  - ☐ d. They are much less reactive than most metals.
15. What are the two most common alkaline earth metals?  
magnesium and calcium
16. Circle the letter of each element that is a transition metal.
- ☒ a. gold                      ☒ b. iron                      ☒ c. copper                      d. lithium
17. Is the following sentence true or false? The transition metals are fairly stable, reacting slowly or not at all with air and water. true

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**R Barnett**



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