



**PHYSICAL SCIENCES (CHEMISTRY)  
GRADE 11 TERM 1 FORMAL EXPERIMENT - PART 2  
VERIFICATION OF BOYLE'S LAW  
MARKING GUIDELINES**

- 1.1 • Aim stated correctly. ✓  
e.g. To determine the relationship between the volume of a fixed amount of gas and the pressure on the gas when temperature remains constant.

(1)

- 1.2 • All apparatus listed ✓

1. Boyle's Apparatus
2. Pump

- All steps of the method correct and logically stated. ✓✓ correct tense used. ✓✓

e.g.

1. The tap on the Boyle's Law apparatus was opened and then the pump was connected.
2. The apparatus was pumped to the pressure.
3. The oil was allowed to settle for a few minutes.
4. A table was used to record the volume reading on the tube and the corresponding pressure reading on the Bourdon pressure gauge.
5. The air tap was slightly opened and quickly closed to let a little air escape from the system. The oil was allowed to settle for a few minutes and then the pressure and volume readings were taken.
6. 5 different readings were recorded in the table.

(5)

**2. Observations and Recording of results:**

Learners will use results obtained during experiment to complete the table below.

(6)

| Volume (V) cm <sup>3</sup> | Pressure (p) kPa |  | pV (°) |
|----------------------------|------------------|--|--------|
|                            |                  |  |        |
|                            |                  |  |        |
|                            |                  |  |        |
|                            |                  |  |        |

Marking guidelines:

- ✓ for correct values of the volume
- ✓ for correct values of the pressure
- ✓✓ for correct values of

# Formal Tests Grade 11

**JE Gale**



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