

AQA Chemistry

GCSE Exam-style mark scheme

C4 Chemical calculations

100 marks (20%)

1 hour

Question	Answer	Mark value (marks)	Mark	Mark
10.1.1	1.4 g		1	100
10.1.2	1.48 g	<p>Amount of magnesium carbonate $1.48 \text{ g} \div 142.04 \text{ g mol}^{-1}$ $= 0.01042 \text{ mol}$ $\times 1000 = 10.42 \text{ mmol}$ <u>10</u> mol as amount of Ca^{2+} ions Bromine ions should be provided by 2 moles</p>	10	100
10.1.3	<p>any 1000 factor</p> <ul style="list-style-type: none"> weighing errors lost or incomplete reaction size of magnesium oxide size of magnesium oxide 		10	100
10.2	<p>all calculations must be shown calculate all ions in the same compound as well as the whole mass of the whole compound</p>	<p>show the formulae of both compounds show the formulae of both ions in the compound show the formulae of both ions in the compound</p>	10	100
10.3	<p>all calculations must be shown</p>		10	100
10.4	<p>all calculations must be shown</p>		10	100
10.5	<p>all calculations must be shown</p>		10	100

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VM Jensen



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web step 7 8 place in an oven preheated to 175 c and bake for 70 minutes take care to place a tray on a lower shelf under the crostata to catch any drippings halfway through take the crostata out of the oven and pierce the holes with a knife once again then return to the oven to bake step 8 8

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