

Mark Scheme

Electrolysis – Molten

1. Ions cannot move
2. Anode (Remember PANIC: Positive Anode Negative Is Cathode).
3. Elements
4. Electrolyte
5. OILRIG - Oxidation Is Loss, Reduction Is Gain (of electrons).
6. Oxidation
7. Ca^{2+} would go to the cathode, I^- would go to the anode.
8. $\text{Ca}^{2+} + 2\text{e}^- \rightarrow \text{Ca}$
9. $2\text{Cl}^- \rightarrow \text{Cl}_2 + 2\text{e}^-$
10. $\text{CaCl}_2 \rightarrow \text{Ca}^{2+} + 2\text{Cl}^-$

Electrolysis – Solutions

1. NaCl
2. Na^+ , H^+ , OH^- , Cl^-
3. Hydrogen, chlorine, sodium hydroxide.
4. Negative ions (OH^- , Cl^-)
5. Positive Ions (H^+ , Na^+)
6. Chlorine, $2\text{Cl}^- \rightarrow \text{Cl}_2(\text{g}) + 2\text{e}^-$
7. Hydrogen, $2\text{H}^+ + 2\text{e}^- \rightarrow \text{H}_2(\text{g})$
8. Na^+ and OH^-
9. Sodium hydroxide
10. Aluminium. Oxygen is released at the anode where it reacts with the graphite to form carbon dioxide. Therefore the anode needs to be replaced often.

Isa Electrolysis Mark Scheme

**Thirumalaisamy P.
Velavancorresponding**



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web acrostics are a fun poetic form that anyone can write they have just a few simple rules and this lesson will teach you how to create acrostic poems of your own to begin with an acrostic is a poem in which the first letters of each line spell out a word or phrase the word or phrase can be a name a thing or whatever you like when children

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web sep 15 2023 write the poem an acrostic is a very special kind of poem in an acrostic the first letters of each line spell out a word or phrase that relates to the topic or theme of the poem they are some of the simplest poems to write but that doesn t make them any less meaningful