

## Identifying Reaction Types & Balancing Equations

Step 1: Name the type of chemical reaction	Step 2: Balance the following chemical equations:
Ex. Combination Reaction	$2 \text{ N}_2 + 6 \text{ H}_2 \rightarrow 4 \text{ NH}_3$
1.	$\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$
2.	$\text{NaCl} + \text{F}_2 \rightarrow \text{NaF} + \text{Cl}_2$
3.	$\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$
4.	$\text{AgNO}_3 + \text{MgCl}_2 \rightarrow \text{AgCl} + \text{Mg}(\text{NO}_3)_2$
5.	$\text{AlBr}_3 + \text{K}_2\text{SO}_4 \rightarrow \text{KBr} + \text{Al}_2(\text{SO}_4)_3$
6.	$\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
7.	$\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
8.	$\text{FeCl}_3 + \text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + \text{NaCl}$
9.	$\text{P} + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$
10.	$\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$
11.	$\text{Ag}_2\text{O} \rightarrow \text{Ag} + \text{O}_2$
12.	$\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$
13.	$\text{K} + \text{MgBr}_2 \rightarrow \text{KBr} + \text{Mg}$
14.	$\text{HNO}_3(\text{aq}) + \text{Ba}(\text{OH})_2(\text{aq}) \rightarrow \text{Ba}(\text{NO}_3)_2(\text{aq}) + \text{H}_2\text{O}(\text{l})$
15.	$\text{C}_3\text{H}_{12}(\text{g}) + \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g})$
16.	$\text{Al}(\text{s}) + \text{Fe}_2\text{O}_3(\text{aq}) \rightarrow \text{Al}_2\text{O}_3(\text{aq}) + \text{Fe}(\text{s})$
17.	$\text{Al}(\text{s}) + \text{O}_2(\text{g}) \rightarrow \text{Al}_2\text{O}_3(\text{s})$
18.	$\text{P}_4 + \text{O}_2 \rightarrow 2 \text{ P}_2\text{O}_5$
19.	$2 \text{ CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
20.	$\text{H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O}$

# Identifying Reaction Types And Balancing Chemical Equations

**E Durkheim**



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**Reaction Green Metrics** John Andraos,2018-12-07 This book contains a series of exercises and problems posed in the subject of green metrics Essentially it is a how to book on evaluating the material efficiency environmental impact safety hazard impact and energy efficiency of any kind of chemical reaction or synthesis plan Only the essential green metrics in each of these categories are used The introduction highlights the hierarchy of metrics used throughout the book explains the structure of how the book is arranged how the problems are posed and how the reader is to use the book Examples refer to themes according to the headings given in the table of contents and are arranged in a hierarchical order Key Features The topics cover fundamentals in chemistry and the chemical industry in a blended fashion A unique text covering the fundamentals of green metrics from materials efficiency and environmental and safety hazard impact to new green technologies and more The book will be useful in a range of chemistry courses from early undergraduate to advanced graduate courses whether based in lectures tutorials or laboratory experiments Using an extensive glossary of terms used in green metrics each chapter has a specified theme where the relevant metrics definitions pertaining to that theme will be given with one or two illustrative worked examples Supplemental web based downloadable material including extra problems full solutions Excel files ChemDraw files templates and exercises

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*Chemistry and Chemical Reactivity* Mary L. Kotz, 1996

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new and updated sections in drilling and production Covers all calculations tables and equations for every day petroleum engineers Features new sections on today s unconventional resources and reservoirs **Holt Chemistry** Ralph Thomas Myers,2004 ICSE-The Science Orbit(Chem)-TB-08-R Rajalaxmi K, Dr R L Madan Former Principal of Government school has put all his expertise and experience in creating these books The books draw immensely from his in depth knowledge and passion for the subject **Management of Water Resources in Protected Areas** Hermes Farfán González,José Luis Corvea Porras,Irene de Bustamente Gutiérrez,James W. LaMoreaux,2013-03-14 Natural ecosystems are heavily dependent on water as it is essential to the development of life The ecology and landscape play an important role in the quality and availability of water It is no coincidence that exceptional hydrological phenomena are found in protected areas Such is the case with for example the geothermic occurrences principally geysers in America s Yellowstone National Park the oldest park in the world The Ramsar wetlands where the ecosystem s dependency on water is strongly evident The Igua u Falls on the border of Argentina and Brazil or the Zapata Swamp the largest of its kind on the Caribbean island of Cuba further exemplify this point However in many cases the conservation strategies for hydraulic resources in protected areas are ignored or simply deprived of the attention they require There are many types of suitable management strategies for planning and protecting our valuable treasures Hydraulic resource management in protected areas is something that must not be separated from these conservation measures The first Symposium for the Management of Hydraulic Resources in Protected Areas was intended to be a framework of communication about experiences with water resource management in protected areas Advances in research and possible solutions to the problems within these areas were discussed The contributions in this proceedings volume are grouped under seven main themes Purification and reuse of wastewater in rural communities Impact of public use on water resources Vulnerability and risks associated with aquifers Design and management water resources in protected areas Research and monitoring of water resources in protected areas Water and its importance as a source of renewable energy in protected spaces and Geodiversity and conservation of areas with hydraulic heritage

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