

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পটীকার প্রস্তুতকরণ

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

$$W = 20 \text{ g}$$

$$HCl \text{ এর আপেক্ষিক ভর,}$$

$$M = 36.5 \text{ g/mol}$$

$$\text{চলকের আয়তন, } V = 1 \text{ L}$$

$$\text{চলকের ঘনমাত্রা } S = 1$$

উদাহরণে উল্লিখিত X-পদ্ধতিতে 10 mg এর মধ্যে কয় পরিমাণ ঘনমাত্রা 0.5 g 2% (W/V) HCl নমুনা নিয়ে কাজ করা হয়েছে।

যদিও X-পদ্ধতিটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও Y-পদ্ধতিটি হলো মাইক্রো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 মিলিগ্রাম রাসায়নিক উপাদান সন্নিবেশিত ও বিশ্লেষণের বিশ্লেষণ ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H₂S এর পরিমাণে কয়েকমিলিগ্রামের ব্যবহার করা হয় ফলে পরিবেশে মুদ্রণ ঘটে না।



অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়।

ফলে পরিবেশে মুদ্রণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

চিত্র-১০



১. উদাহরণে দ্রব্যের ভী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঞ্জার প্রতীকটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহরণের অংশের চলক থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার চলক তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪ নং প্রশ্নের উত্তর

১. বিশ্লেষণের নিয়ন্ত্রিত কক্ষের উপর রাইজার স্থাপন করলে কক্ষের প্রতি দিকের দৃশ্য যে ভর পাওয়া যায়, তাকে রাইজার দ্রব্যক বলে।

২. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 মিলিগ্রাম রাসায়নিক উপাদান সন্নিবেশিত ও বিশ্লেষণের বিশ্লেষণ ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

৩. অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মুদ্রণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

৪. উদাহরণের চলকটি হলো সালফিউরিক এসিড (H₂SO₄) এর 10 M চলক। H₂SO₄ একটি জারক ও ক্ষয়কারী পদার্থ। H₂SO₄ এর 10 M চলক হলে পাত্র H₂SO₄ চলক। পাত্র H₂SO₄ পদার্থকে ক্ষয়কারী হ্যাঞ্জার প্রতীক অঙ্কন করা হয়। পাত্র H₂SO₄ এর জন্য ক্ষয়কারী হ্যাঞ্জার প্রতীক নিয়ে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রতীক	ক্ষয়কারী (Corrosive)
আয়তন	ক্ষয়কারী রাসায়নিক উপাদান চলকে যে ক্ষয়কারী রাসায়নিক দ্রব্যকে মুদ্রণের সময় সংশ্লিষ্ট তরল ও অন্যান্য পদার্থের মধ্যস্থিত ক্ষয়কারী ক্ষতি সাধিত হয়। এতে তরল পুড়ে যায় এবং অন্যান্য পদার্থ পুড়ে যায়।
উদাহরণ	পাত্র কঠিন এসিডসমূহ (HCl, HNO ₃ , H ₂ SO ₄), ক্ষয়কারী (NaOH, KOH), H ₂ O ₂ , অক্সিজেন লিকার প্রভৃতি চলক ইত্যাদি।
সংরক্ষণ পদ্ধতি	- পাত্র এসিডকে পোতলে রাখতে হবে। - পাত্র ক্ষয়কারী, ক্ষয়কারী পদার্থের পিছনে রাখতে হবে।

উদাহরণের চলকটি হলো 10M H₂SO₄ চলক এবং এর থেকে 1M ঘনমাত্রার চলক প্রস্তুত করতে হবে 1ম ও ২য় চলকের ব্যবহার করতে হবে।

1ম চলকটি হলো 100 ml অক্সিজেনমুক্ত চলক ও এর 10 ml পিপিটে। যেহেতু অক্সিজেনমুক্ত চলকের আয়তন 100 ml, তাই আমাদের 100 ml 1M ঘনমাত্রার H₂SO₄ চলক তৈরি করতে হবে। চলকের প্রস্তুতকরণ পুত্র হতে জানতে পারি,

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

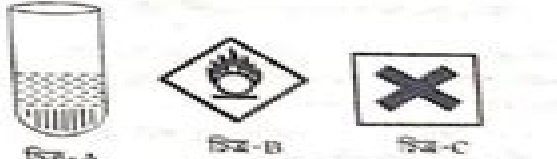
$$= 10 \text{ mL}$$

অতএব, 10ml H₂SO₄ চলকের সঙ্গে পানি যোগ করতে হবে

$$(100 - 10) \text{ mL} = 90 \text{ mL}$$

অতএব, পিপিটের সাহায্যে 10 ml প্রথম ঘনমাত্রার চলক নিয়ে তার সাথে অক্সিজেনমুক্ত চলকে 90 ml পানি যোগ করলে আমাদের কাঙ্ক্ষিত ঘনমাত্রার চলক পাওয়া যাবে।

চিত্র-১১



১. অক্সিজেনমুক্ত করে নেওয়া
২. সামান্য দ্রব্য K₂ এর ঘন কখনো অসীম হয় না-ব্যাখ্যা করে।
৩. A চিহ্নের হ্যাঞ্জারের উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C চিহ্নের সাথে সম্পর্কিত রাসায়নিক দ্রব্যটির ব্যবহারকালে খান্সা কৃতি হ্রাসে কবীর উপর সম্পর্ক বিশ্লেষণ করে। ৪ নং প্রশ্নের উত্তর

অক্সিজেনমুক্ত করে নেওয়া রাসায়নিক বা কৃত্রিম দ্রব্য সঞ্চেপ্ত বা মুদ্রিত বা উপাদানসমূহকে জারণ প্রতিরোধ কক্ষ থেকে রক্ষা করে রাখার সঞ্চেপ্ত করে।

৩. সামান্য দ্রব্য K₂ এর ঘন কখনো অসীম বা শূন্য হতে পারে না। উদাহরণ ব্যাখ্যা করার জন্য নিম্নের উদাহরণী মাইক্রোপট বিশ্লেষণ করি-

$$A + B \rightarrow C + D$$

Hsc Chemistry 1 First Chapter Notes

Kathleen Armour



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