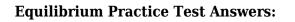
#### Equilibrium Practice Test

reservation (	Name: ANSWER	.5	60 /60
obliple Choice outify the letter of the choice that h	est completes the states	most or assesses the c	guestion. (20 marks)
I. Equilibrium hoppens when			
a, opposing forces bulance	4.1	the forward and rever	rise reactions occur at different rates
b. opposing forces are unbal-	anced d.	the forward and rever	rse reactions are unbalanced
2. Which is a macroscopic prog	ocave?		
a, colour	40.7	conscinutions.	
b. temperature	dL.	pHI	
3. A system reaches equilibrius	m. Which must be true	e of the system?	
a. it is open		it is either open or cle	
b. It is closed	d.	it is open and has pre-	"ANUES"
	son of water with a lid	on it. Why does this	s situation <u>not</u> represent an equilibrius
system?			
<ul> <li>a. It is not closed system.</li> <li>b. The equilibrium cannot be</li> </ul>	a contract of the contract of	and the second second	
c. The temperature is not con		pperson disposion.	
d. The rates of opposing ch			
E. Millert, of these to NOVE on the	Control of the second control of	and the second second second	
<ol> <li>Which of these is <u>NOT</u> an ex a. H<sub>2</sub>O(1)<sub>000</sub>H<sub>2</sub>O(g)</li> </ol>		N <sub>2</sub> O <sub>2</sub> (g) as 2NO <sub>2</sub> (g)	
b. C(s) + CO <sub>2</sub> (g)e+CO <sub>2</sub> (g)		C:H-OH(Des-C:H-O)	Hilleri
			_
<ol><li>In the equilibrium constant</li></ol>	t expression below, wi	set do the expensions	s represent?
$K_{eq} = \frac{(C)^{n} (D)^{n}}{(A)^{n} (D)^{n}}$			
a the concentrations of react b, the equilibrium constant			ed and reverse reactions settletents from the balanced equation
7. Which is tree of the permeric	out waters of the condition	orthographene annutra	
a. It represents the relative		The state of the s	
b. It represents the ratio of re-			
c. It is measured in the SI un	it, mole.		
d. Both a and c are true.			
8. What does it mean if $\mathbb{K}_{+*} < 1$	at confillerium?		
a. The concentration of rea		the concentration of	f products.
b. The concentration of prod			
c. The rate of the forward re-			
d. The rate of the reverse rea	ction is greater than the	ritate of the forward r	traction.
9. When the equilibrium const the initial concentr		approximate concer	stration of reactant at equilibrium is
as greater than		the same on	
b. less than	d.	exactly 100 times	
10. If Q <sub>ac</sub> > K <sub>af</sub> , then			
a, the system is at equilibrium	_	of the matter of t	products to reactants is greater than 1
b. the ratio of products to rea			shifts toward product formation
11. Use the following informati-			_
The following reaction is an H <sub>2</sub> (g) + L <sub>2</sub> (g) so 2HI(g)	example of homogeneo	sus equilibrium.	
	doubling the volume	and the temperatur	re remains constant, the equilibrium w
a. shift forward.		not be affected	
b. shoth backward	d.	will not reach equilib	eriumo.

# **Equilibrium Practice Test Answers**

**TD Snyder** 



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