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Electroanalytical Chemistry Principles Best Practices and Case Studies begins by introducing some basic concepts in electrical phenomena It then moves on to a chapter that examines the potentiometry of oxidation reduction processes followed by another on the potentiometry of ion selective electrodes Other sections look at applications of ion selective electrodes controlled potential methods case studies in controlled potential methods and instrumentation The book also features several appendixes covering Ionic Strength Activity and Activity Coefficients The Nicolsky Eisenman Equation The Henderson Equation for Liquid Junction Potentials Selected Standard Electrode Potentials and The Nernst Equation Derivation Introduces the principles of modern electrochemical sensors and instrumental chemical analysis using potentiometric and voltammetric methods Develops conceptual models underlying electrochemical phenomena and useful equations Illustrates best practice with short case studies of organic reaction mechanisms using voltammetry and quantitative analysis with ion selective electrodes Offers instructors the opportunity to select focus areas and tailor the book to their course by providing a collection of shorter texts each dedicated to a single field Intended as one of a series of modules for teaching undergraduate courses in instrumental chemical analysis Electroanalytical Chemistry Principles Best

Practices and Case Studies is an ideal textbook for undergraduate majors in chemistry and chemical engineering taking instrumental analysis courses It would also benefit professional chemists who need an introduction to potentiometry or voltammetry *A Manual of the Principal Instruments Used in American Engineering and Surveying* W. & L.E. Gurley,1895

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