

- (c) An atomic nucleus A is composed of 84 protons and 128 neutrons.
- The nucleus A emits an alpha particle and is transformed into nucleus B. What is the composition of nucleus B?
 - The nucleus B emits a beta particle and is transformed into a nucleus C. What is the composition of nucleus C?
 - Does the composition of nucleus C change if it emits gamma radiations?

[3]

Examiners' Comments

- Most candidates failed to understand that the thermionic emission is due to the heat energy being imparted to the metallic surface. In place of an electron volt candidates wrote joule.
- Many candidates drew incorrect diagrams with either the radiations being labelled incorrectly or the deflection shown incorrectly.
- Most candidates did not understand the changes coming in with the emission of an alpha particle or a beta particle. Many failed to write the composition of the nucleus or explain appropriately the number of protons and neutrons present in the nucleus after the emission with a few unable to understand the effect of Gamma emissions.

Suggestions for teachers

- Explain the difference between thermionic and electronic emission.
- The unit of work function and its relation with the S.I unit of the physical quantity which it measures can be explained at this stage.
- Comparative property of radioactive radiations in an electric field should be discussed during classroom teaching.
- Stress must also be laid to teach the atomic structure in the class so as to make it clearer.

MARKING SCHEME

Question 10.

- Emission of free electrons from a metal surface when heat energy is imparted to it is called the thermionic emission.
 - Electron Volt.
- Deflection of Alpha and Beta shown correctly. Alpha deflection is less than beta Path of Gamma.
 - Use of lead lined aprons/ Lead gloves / long handle lead tongs/ radioactive substance should be kept in a thick walled lead container (Any two)
- ${}_{82}\text{B}^{208}$ or 82 protons and 126 neutrons
 - ${}_{83}\text{C}^{208}$ or 83 protons and 125 neutrons
 - No.

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M Carnoy



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Table of Contents Icse 2015 Physics Question Paper

1. Understanding the eBook Icse 2015 Physics Question Paper
 - The Rise of Digital Reading Icse 2015 Physics Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Icse 2015 Physics Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Icse 2015 Physics Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Icse 2015 Physics Question Paper
 - Personalized Recommendations
 - Icse 2015 Physics Question Paper User Reviews and Ratings
 - Icse 2015 Physics Question Paper and Bestseller Lists

5. Accessing Icse 2015 Physics Question Paper Free and Paid eBooks
 - Icse 2015 Physics Question Paper Public Domain eBooks
 - Icse 2015 Physics Question Paper eBook Subscription Services
 - Icse 2015 Physics Question Paper Budget-Friendly Options
6. Navigating Icse 2015 Physics Question Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Icse 2015 Physics Question Paper Compatibility with Devices
 - Icse 2015 Physics Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Icse 2015 Physics Question Paper
 - Highlighting and Note-Taking Icse 2015 Physics Question Paper
 - Interactive Elements Icse 2015 Physics Question Paper
8. Staying Engaged with Icse 2015 Physics Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Icse 2015 Physics Question Paper
9. Balancing eBooks and Physical Books Icse 2015 Physics Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Icse 2015 Physics Question Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Icse 2015 Physics Question Paper
 - Setting Reading Goals Icse 2015 Physics Question Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Icse 2015 Physics Question Paper
 - Fact-Checking eBook Content of Icse 2015 Physics Question Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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