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Puja Mehta



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web step 4 the reactant that produces a smaller amount of product is the limiting reagent mg mg produces less mgo mgo than does o₂ o₂ 3 98 g mgo vs 25 2 g mgo therefore mg is the limiting reagent in this reaction step 5 the reactant that produces a larger amount of product is the excess reagent

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web if you start with 14 82 g of ca oh₂ c a o h₂ and 16 35 g of h₂so₄ h₂ s o₄ a determine the limiting reagent b determine the number of moles of h₂o h₂ o produced c determine the number of grams of caso₄ c a s o₄ produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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