HYDROCARBONS



Hydrocarbon Compounds

Hydrocarbon compounds are compounds that consist only of carbon and hydrogen atoms. Hydrocarbons can be divided into 2, namely:

a. Saturated Hydrocarbons: single bonds

including saturated hydrocarbons:

- Alkane: open chain (C_nH_{2n+2})
- a Cycloalkane : closed chain (C_nH_{≥n})

b. Unsaturated Hydrocarbons: double bonds

- Alkene : double bond (C_nH_{2n})
- Alkyne : triple bond (C_nH_{2n-2})
- Alkadiene : two double bonds (C_nH_{2n-2})



Nomenclature of Alkanes

The nomenclature of alkanes is the basis for the nomenclature of other carbon compounds. Therefore the nomenclature of alkanes must be thoroughly understood.

Here are the names of askanes from C1-C10:

Formula	Name	Formula	Name
CH4	methane	C6H14	hexane
C ₂ H ₆	ethane	C7H16	heptane
C ₃ H ₈	propane	C ₈ H ₁₈	octane
C4H10	butane	C9H20	nonane
C ₅ H ₁₂	pentane	C ₁₀ H ₂₂	decane

Atkene names :

 C_2H_4 = ethylene

 C_3H_6 = propene

Hydrocarbons Note Taking Guide

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Hydrocarbons Note Taking Guide Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Hydrocarbons Note Taking Guide**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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