$pH = -log_{10}[H^*]$

 $\frac{1}{2} F_{\alpha}(g) + g^{\alpha}$

Chemistry

DATA SHEET

Avogadro constant, NA		6.022	$\times 10^{23} \text{ mol}^{-1}$
Volume of I mole ideal gas: at			
	at 0°C (273.15 K)	22.71	L.
	at 25°C (298.15 K)	24.79	L
	25°C (298.15 K), K _w		
Specific heat capacity of water		$4.18 \times$	c 10 ³ J kg ⁻¹ K ⁻¹

Some useful formulae

Some standard potentials

 $\Delta H = -m C \Delta T$

2.89 W

AND DESCRIPTION OF		ar o poetarears	
K* + e*	apple.	BCGr0	-2.94 W
Ba2+ + 2e-	ng-Mr.	Ba(x)	-2.91 W
Ca2+ + 2e-	mark.	Ca(x)	-2.87 W
Na* + e*	negation .	Nia(a)	-2.71 V
Mg2+ + 2e"	ngoliti .	Mg(x)	-2.36 V
All * 3e*	spids.	Allon	-1.68 V
Mas2+ 2e	with:	Mim(x)	-1.18 V
H ₂ O + e ⁻	age to	$\frac{1}{2}H_2(g) + OH^-$	-0.83 V
$2m^{2a} + 2e^{-}$	with:	Zim(x)	-0.76 V
Fe ³⁺ + 2e ⁻	quite.	Plet(x)	-0.44 V
No ²⁺ = 2e ⁻	1976	Niloto	-0.24 V
Ser ²⁺ + 2e ⁻	Spide.	Sin(x)	-0.14 W
Pb ²⁺ + 2e ⁻	spile.	P%(x)	-0.13 V
H* + e*	age the	$\frac{1}{2} H t_2(x)$	0.00 V
$5O_4^{2-} = 4H^* + 2e^*$	ngs/h	$SO_{y}(mq) = 2H_{y}O$	0.16 V
Cu ^{‡=} + 2e	reprint.	Checks	0.34 W
$\frac{1}{2}O_2(g) + H_2O + 2e^{-g}$	nests.	2OH-	0.40 V
Cu* + e*	spells.	Clocks	0.52 V
$\frac{1}{2}I_2(x) = e^{-x}$	ngoliti.	1	0.54 W
$\frac{1}{2}I_{g}(\alpha q) + e^{-}$	aprile .	125	0.62 V
Fe ³⁺ - e ⁻	spile	Fe ³ *	0.77 V
Ag" - e"	reprint.	AgO	0.80 V
$\frac{1}{2} Be_2(t) = e^{-t}$	negative.	19 ₄	1.08 V
$\frac{1}{2}\operatorname{Be}_2(\operatorname{ag}) + e^-$	1996	Br	1.10 W
$\frac{1}{2}O_2(g) + 2H^* + 2e^-$	Special Control	H ₂ O	1.23 W
$\frac{1}{2}Cl_2(g) + g^*$	400	CI	1.36 W
$\frac{1}{2}Ce_2O_2^{2-} + 7H^* + 3e^-$	apple.	$Cr^{3-} = \frac{2}{2}H_2O$	1.36 W
Cl ₂ (mq) + e	spik.	CIT	1.40 W
MeO ₄ " + 8H" + 5e"	age the	$Mm^{2+} + 4H_dO$	1.51 W

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Jianli Hu, Dushyant Shekhawat

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