

Chemistry Unit 8 Worksheet 4

Examples of Every Kind of Problem

This is a separate sheet of paper, so this is a complete solution to each of the problems. Notice: I'll state the problems without an arrow. We have to write your final answer.

12. Calculate the number of moles of potassium chlorate, KClO_3 , that must decompose to produce potassium chloride KCl and a total of 1.5 moles of oxygen gas.



Subst:	1.0	0	0
Change:	-1.0	+1.0	+1.5
Final:	0	1.0	1.5

1.0 moles KClO_3 needed

13. As a single-displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. When a strip of metal is placed in acid, bubbles are immediately seen with a large amount of hydrogen?



Subst:	0.100	0	0	0
Change:	-0.100	-0.200	+0.100	+0.100
Final:	0	0	0.100	0.100

0.100 moles H_2 produced

14. Ethane, C_2H_6 , reacts with oxygen gas to produce carbon dioxide gas and water vapor. What number of moles of oxygen gas is required to react with 0.50 moles of ethane?



Subst:	0.500	0.00	0	0
Change:	-0.500	-0.750	+1.000	+1.500
Final:	0	0.250	1.000	1.500

0.75 moles O_2 needed (0.50 moles $\times$ 1.5) or **0.25 moles CO_2 produced**

15. Sodium metal that is melted reacts with chlorine gas to produce sodium chloride. 10.0 g of melted Na reacts, reacted with a sufficient amount of chlorine gas to produce the following equation:



Subst:	0.430	0.00	0
Change:	-0.430	-0.215	+0.430
Final:	0	0.215	0.430

0.215 moles Cl_2 needed or **0.430 moles NaCl produced**

Modeling Chemistry U8 Ws 3 V2 Key

L Cohen



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