

Mixed Gas Laws Worksheet

- 1) How many moles of gas occupy 98 L at a pressure of 2.8 atmospheres and a temperature of 292 K?

$$V = 98 \text{ L}$$

$$P = 2.8 \text{ atm}$$

$$T = 292 \text{ K}$$

$$n = ?$$

$$R = 0.082 \frac{\text{L atm}}{\text{mol K}}$$

$$PV = nRT$$

$$n = \frac{PV}{RT}$$

$$= \frac{2.8 \times 98}{0.082 \times 292} = \underline{\underline{11.46 \text{ mole}}}$$

- 2) If 5.0 moles of O_2 and 3.0 moles of N_2 are placed in a 30.0 L tank at a temperature of 25°C , what will the pressure of the resulting mixture of gases be?

$$5 \text{ moles } \text{O}_2$$

$$3 \text{ moles } \text{N}_2$$

$$V = 30 \text{ L}$$

$$T = 25^\circ \text{C} \rightarrow 273 + 25$$

$$P = ?$$

$$PV = nRT$$

$$P = \frac{nRT}{V} = \frac{8 \times 0.082 \times 298}{30 \text{ L}}$$

$$= \underline{\underline{6.5 \text{ atm}}}$$

- 3) A balloon is filled with 35.0 L of helium in the morning when the temperature is 20.0°C . By noon the temperature has risen to 45.0°C . What is the new volume of the balloon?

$$① V_1 = 35 \text{ L}$$

$$① T_1 = 20^\circ \text{C} = 293 \text{ K}$$

$$② T_2 = 45^\circ \text{C} = 318 \text{ K}$$

$$② V_2 = ?$$

Charles law

$$\frac{V_1}{T_1} = \frac{V_2}{T_2}$$

$$\frac{35}{20} = \frac{V_2}{45}$$

$$V_2 = \frac{35 \times 318}{293} = \underline{\underline{37.78 \text{ L}}}$$

- 4) A 35 L tank of oxygen is at 315 K with an internal pressure of 190 atmospheres. How many moles of gas does the tank contain?

$$V = 35 \text{ L}$$

$$T = 315 \text{ K}$$

$$P = 190 \text{ atm}$$

$$n = ?$$

$$PV = nRT$$

$$n = \frac{PV}{RT} = \frac{190 \times 35}{0.082 \times 315}$$

$$= \underline{\underline{257.45 \text{ mol}}}$$

Mixed Gas Law Worksheet Answers

Scott C. Dulebohn

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

Mixed Gas Law Worksheet Answers:

Reviewing **Mixed Gas Law Worksheet Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Mixed Gas Law Worksheet Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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