CHAPTER 7 PRACTICE TEST: ATOMIC STRUCTURE & PERIODIC TRENDS- KEY

	odium chloride ow C) Gree	is vaporized in a flame, the n D) Violet E) V		
	liation travels th	rough vacuum at a speed	of m/s.	
	C) 3.00×10 ⁸		1	
B) 125	D) 10,000	λ		
3) The wavelength of	light that has a f	requency of 1.20×10 ^{t3} s ⁻¹	is m.	
A) 25.0	B) 2.50×10 ⁻⁵		E) 2.5	
	The state of the s	D) 12.0	5186-16106-0	
GIVEN: $\nu = 1.20 \times 10$ CONNECTION: $c = 3$		JNKNOWN; λ		
SUBSTITUTE & SOLVE:				
$\lambda = 3.00 \times 10^8 \text{ m} \cdot \text{s}^{-1}$ $1.20 \times 10^{13} \text{ s}^{-1}$				
$\lambda = 2.50 \times 10^{-5} \mathrm{m}$				
			ne frequency of this EM ra	
A) 500	B) 200		D) 20	E) 2.0
GIVEN: $\lambda = 6 \text{ m}$ CONNECTION: $c = \lambda$		OWN: v (Must convert ans	wer to MHz!)	
SUBSTITUTE & SOLVE:				
$\nu = \frac{3.00 \times 10^8 \text{ m} \cdot \text{s}^{-1}}{6 \text{ m}}$				
≥ = 0.5 x 10 ⁸ Hz x	10° Hz =	50 MHz		
5) The energy of a pho its wavelength.	oton of light is _	proportional to	o its frequency and	proportional to
A) directly, directly		C) inversely, directly	E) indirect	tly not
B) inversely, inversely		D) directly, inversely	all many	11, 100
		avelength of 12.3 nm is _	J.	
A) 1.51×10 ⁻¹⁷	B) 4.42×10 ⁻²³	C) 1.99×10 ⁻²⁸	D) 2.72×10 ⁻⁵⁰	E) 1.62×10 ⁻¹⁷
7) The wavelength of	a photon that ha	s an energy of 5.25×10 ⁻¹⁹	J is m.	
A) 3.79×10 ⁻⁷	B) 2.64×10°	C) 2.38×10 ²³	D) 4.21×10 ⁻²⁴	E) 3.79×10 ⁷
8) A mole of red photo	ons of waveleng	th 725 nm has	kJ of energy.	
A) 2.74×10 ⁻¹⁹	B) 4.56×10-4	C) 6.05×10 ⁻³	D) 165	E) 227
GIVENS: $\lambda = 725$ nm s $\lambda = 725 \text{ nm} \text{ x}$	must convert to	m before using formulas		

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