

Sample Problem 1 Worked Out



Step 1 Find the limiting reagent:

$$\begin{array}{ccccccc} & 1000 \text{ g H}_2 & 1 \text{ mol H}_2 & 2 \text{ mol NH}_3 & & & \\ 5.22 \text{ kg H}_2 & \times \frac{\text{-----}}{1 \text{ kg H}_2} & \times \frac{\text{-----}}{2.02 \text{ g H}_2} & \times \frac{\text{-----}}{3 \text{ mol H}_2} & = & 1.72 \times 10^3 \text{ mol NH}_3 \end{array}$$

$$\begin{array}{ccccccc} & 1000 \text{ g N}_2 & 1 \text{ mol N}_2 & 2 \text{ mol NH}_3 & & & \\ 31.5 \text{ kg N}_2 & \times \frac{\text{-----}}{1 \text{ kg N}_2} & \times \frac{\text{-----}}{28.02 \text{ g N}_2} & \times \frac{\text{-----}}{1 \text{ mol N}_2} & = & 2.25 \times 10^3 \text{ mol NH}_3 \end{array}$$

H₂ is the limiting reagent

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