

## Limiting Reagent and Percentage Yield

- 1 Dichlorine monoxide is produced by passing chlorine gas over heated mercury (II) oxide according to the following unbalanced reaction:



Balance the reaction. Suppose the quantity of the reactants is sufficient to produce 0.86 g of  $\text{Cl}_2\text{O}$ , but only 0.71 g is obtained. What is the percentage yield?

- 2 Consider the following reaction:



The theoretical yield of beryllium chloride was 10.7 g. Suppose the reaction actually yields 4.5 g. What is the percentage yield?

- 3 Nitrogen and hydrogen at high temperatures are converted to ammonia using the following reaction:



When 400 g of  $\text{H}_2$  are added to an excess amount of  $\text{N}_2$ , 104 g of  $\text{NH}_3$  are formed. Calculate the percentage yield.

# Limiting Reactants And Percent Yield Review Sheet

**AN Whitehead**



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reply to the employee within 28 days tell them they re entitled to 52 weeks maternity leave confirm their return to work date they can change it later if they want to take less but must give you 8 weeks notice if the employee tells you how long they want off use this to work out their return to work date

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use this letter when you need to communicate with an employee about their return from maternity leave date and any related arrangements or queries that they may have raised with you the template provides drafting options to cover the following scenarios and topics

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