

- 10 In 1859, Thomas Edison invented the nickel-iron battery. The following reaction takes place in the battery.



What is the limiting reactant when 5.00 mol of Fe and 8.00 mol of  $\text{Ni}(\text{OH})_2$  react?

How many moles of  $\text{Fe}(\text{OH})_2$  will be produced?

How many moles of the other reactant will be in excess?

- 20 One of the first cerium compounds that forms is cerium sesquifluoride ( $\text{Ce}_2\text{CeF}_7$ ).



What is the limiting reactant in the reaction of 12.5 mol of cerium fluoride with 50.0 mol of sesquifluoride?

How many moles of  $\text{Ce}_2\text{CeF}_7$  can be produced?

How many moles of the other reactant will be excess?

- 30 Iron is obtained commercially by the reaction of hematite ( $\text{Fe}_2\text{O}_3$ ) with carbon monoxide.



What is the limiting reactant when 25.0 mol of hematite reacts with 30.0 mol of carbon monoxide?

How many grams of Fe can be produced?

# Limiting Reactants And Percent Yield 9 3 Answer Key

**Allan Blackman, Daniel  
Southam, Gwendolyn Lawrie, Natalie  
Williamson, Christopher Thompson**

### **Limiting Reactants And Percent Yield 9 3 Answer Key:**

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