

Name: _____ Date: _____

Electrons in Atoms Study Worksheet

Fill in the blanks in the following paragraph with the appropriate terms.

In Bohr's model of the atom, electrons inhabit certain _____ levels, with the levels closest to the nucleus having _____ energy than those farther from the nucleus. In the _____ state of the atom, the electrons are in the lowest _____ level possible. When an atom absorbs energy, it is said to ascend to an _____ state, where it is unstable. The atom will soon _____ the same amount of energy absorbed which may be seen in the form of visible light. In the study of _____, this visible light is seen as the _____ spectrum of an element. This spectrum is _____ to each element, hence it is also called an element's "fingerprints". The modern view of light is that it has a _____ nature. In other words, light may behave as a stream of particles called _____ or _____, or light may behave as a _____.

In the wave view of light, the wave equation is often used to determine a wave's frequency or wavelength. The _____ is the distance between corresponding points on adjacent waves while the _____ is the number of waves passing a given point in a given time. The wave equation is: _____.

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