

Chapter 1

Introduction to Electronics

Question A.1: Atomic Structure

- (a) The atom is the smallest constituent of matter that retains all the chemical properties of the element.
- (b) The atom is made up of three subatomic particles: electrons, protons, and neutrons.

Question A.2: Materials Used for Electronics

- (a) The materials used for making electronic components are classified into three categories: conductors, semiconductors, and insulators.
- (b) The most commonly used semiconductor material is silicon.
- (c) A material is said to be a semiconductor if its conductivity lies between that of a conductor and an insulator.

Question A.3: Current in Semiconductors

- (a) Current is the flow of electric charge. In semiconductors, it is carried by both electrons and holes.
- (b) Current is produced when an electric field is applied across a semiconductor material.

Question A.4: N-Type and P-Type Semiconductors

- (a) An N-type semiconductor is formed by adding a small amount of pentavalent impurity to an intrinsic semiconductor. The impurity atoms have five valence electrons, one of which is loosely bound and can move freely, creating free electrons.
- (b) A P-type semiconductor is formed by adding a small amount of trivalent impurity to an intrinsic semiconductor. The impurity atoms have three valence electrons, leaving a vacancy or hole in the valence band, which can move freely.

Question A.5: The PN Junction

- (a) The PN junction is formed by joining an N-type semiconductor with a P-type semiconductor. The interface between them is called the depletion region. In this region, there is a built-in electric field due to the difference in carrier concentrations.
- (b) The PN junction is used in many electronic devices, such as diodes, transistors, and solar cells. It is the basic building block of most modern electronic circuits.

Electronic Devices 9th Edition Solution Manual

Jin-Ying Zhang



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