

## Electron Configuration Practice Worksheet

*In the space below, write the unabbreviated electron configurations of the following elements:*

- 1) sodium \_\_\_\_\_
- 2) iron \_\_\_\_\_
- 3) bromine \_\_\_\_\_
- 4) barium \_\_\_\_\_
- 5) neptunium \_\_\_\_\_

*In the space below, write the abbreviated electron configurations of the following elements:*

- 6) cobalt \_\_\_\_\_
- 7) silver \_\_\_\_\_
- 8) tellurium \_\_\_\_\_
- 9) radium \_\_\_\_\_
- 10) lawrencium \_\_\_\_\_

Determine what elements are denoted by the following electron configurations:

- 11)  $1s^2 2s^2 2p^6 3s^2 3p^4$  \_\_\_\_\_
- 12)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$  \_\_\_\_\_
- 13)  $[\text{Kr}] 5s^2 4d^{10} 5p^3$  \_\_\_\_\_
- 14)  $[\text{Xe}] 6s^2 4f^{14} 5d^6$  \_\_\_\_\_
- 15)  $[\text{Rn}] 7s^2 5f^{11}$  \_\_\_\_\_

Determine which of the following electron configurations are not valid:

- 16)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 4d^{10} 4p^5$  \_\_\_\_\_
- 17)  $1s^2 2s^2 2p^6 3s^3 3d^5$  \_\_\_\_\_
- 18)  $[\text{Ra}] 7s^2 5f^6$  \_\_\_\_\_
- 19)  $[\text{Kr}] 5s^2 4d^{10} 5p^5$  \_\_\_\_\_
- 20)  $[\text{Xe}]$  \_\_\_\_\_

# Electron Configurations Worksheet Pogil

**Xiaolong Qi**



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## **Table of Contents Electron Configurations Worksheet Pogil**

1. Understanding the eBook Electron Configurations Worksheet Pogil
  - The Rise of Digital Reading Electron Configurations Worksheet Pogil
  - Advantages of eBooks Over Traditional Books
2. Identifying Electron Configurations Worksheet Pogil
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Electron Configurations Worksheet Pogil
  - User-Friendly Interface

4. Exploring eBook Recommendations from Electron Configurations Worksheet Pogil
  - Personalized Recommendations
  - Electron Configurations Worksheet Pogil User Reviews and Ratings
  - Electron Configurations Worksheet Pogil and Bestseller Lists
5. Accessing Electron Configurations Worksheet Pogil Free and Paid eBooks
  - Electron Configurations Worksheet Pogil Public Domain eBooks
  - Electron Configurations Worksheet Pogil eBook Subscription Services
  - Electron Configurations Worksheet Pogil Budget-Friendly Options
6. Navigating Electron Configurations Worksheet Pogil eBook Formats
  - ePub, PDF, MOBI, and More
  - Electron Configurations Worksheet Pogil Compatibility with Devices
  - Electron Configurations Worksheet Pogil Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Electron Configurations Worksheet Pogil
  - Highlighting and Note-Taking Electron Configurations Worksheet Pogil
  - Interactive Elements Electron Configurations Worksheet Pogil
8. Staying Engaged with Electron Configurations Worksheet Pogil
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Electron Configurations Worksheet Pogil
9. Balancing eBooks and Physical Books Electron Configurations Worksheet Pogil
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Electron Configurations Worksheet Pogil
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Electron Configurations Worksheet Pogil
  - Setting Reading Goals Electron Configurations Worksheet Pogil
  - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Electron Configurations Worksheet Pogil
  - Fact-Checking eBook Content of Electron Configurations Worksheet Pogil
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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