Electron	Configuration	Practice
Chemistr	y	

Maime:		
Dans Da		

How to write an electron configuration:

- A. Determine the total number of electrons to be represented.
- B. Use the Aufbau principle to fill the orbitals with electrons for elements 1-23. Refer to electron configuration periodic table for elements after 23
- C. The sum of the superscripts should equal the total number of electrons. Example: 12Mg Is<sup>2</sup> 2s<sup>2</sup> 2p<sup>6</sup> 3s<sup>2</sup>

#### I. Configuration Writing Practice

- A. Write a ground state electron configuration for each neutral at: 1. Ground state means that all of the lowest possible energy levels (up to the pr per number of electrons for the element) are filled.
- 1. No: 1522522p6351
- 2. Pb: 1522522p63523p64523d104p65524d105p66524f145d106p2
- 3. Sr: 1522522p63523p64523d104p6552
- 4. U: 1522522p63523p64523d104p65524d105p66524f145d106p67525f36d1
- 5. N: 152 252 2p3
- \*6. Ag: 1522522p63523p64523d104p65514d10
  - 7. Ti: 1522522p6 3523p6 452 3d2
  - 8. Ce: 1522522p63623p64523d104p65524d105p66524f15d1
  - 9. cl: 1522522p63523p5
  - 10. Hg: 1522522p63523p6 4523d104p65524d105p66524f145d10

# **Electron Configurations Packet With Answers**

**LP Steffe** 



This book delves into Electron Configurations Packet With Answers. Electron Configurations Packet With Answers is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Electron Configurations Packet With Answers, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Electron Configurations Packet With Answers
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  - ∘ Chapter 5: Conclusion
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- 6. In chapter 5, this book will draw a conclusion about Electron Configurations Packet With Answers. The final chapter will summarize the key points that have been discussed throughout the book.
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