# Electron Configurations Worksheet - Answers

Write the complete ground state electron configurations for the following:

1)	lithium	1s <sup>2</sup> 2s <sup>1</sup>
2)	oxygen	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>4</sup>
3)	calcium	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup>
4)	titanium	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 3d <sup>2</sup>
5)	rubidium	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>6</sup> 5s <sup>1</sup>
6)	lead	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>6</sup> 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>6</sup> 6s <sup>2</sup> 4f <sup>14</sup> 5d <sup>10</sup> 6p <sup>2</sup>
7)	erbium	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 4s <sup>2</sup> 3d <sup>10</sup> 4p <sup>6</sup> 5s <sup>2</sup> 4d <sup>10</sup> 5p <sup>6</sup> 6s <sup>2</sup> 4f <sup>12</sup>

Write the abbreviated ground state electron configurations for the following:

8)	helium	1s <sup>2</sup> (this one cannot be abbreviated)	
9)	nitrogen	[He] 2s <sup>2</sup> 2p <sup>3</sup>	
10)	chlorine	[Ne] 3s <sup>2</sup> 3p <sup>5</sup>	
11)	iron	[Ar] 4s <sup>2</sup> 3d <sup>6</sup>	
12)	zinc	[Ar] 4s <sup>2</sup> 3d <sup>10</sup>	
13)	barium	[Xe] 6s <sup>2</sup>	
14)	polonium	[Xe] 6s <sup>2</sup> 4f <sup>14</sup> 5d <sup>10</sup> 6p <sup>4</sup>	

# **Electron Configuration Supplement Worksheet Answers**

**VM** Jensen

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