Name:	Date:	
Score:		

# H

		ection coi	iliguration	
L. Determine wha	at elements a	ere denoted by t	he following confid	aurations.
1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3				
i. 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup>		217		
ii. 1s² 2s² 2p <sup>6</sup> 3s²				
		450 Eco		
v. [Kr] 5s <sup>2</sup> 4d <sup>10</sup> 5p	)-"	-		
. [Ne] 3s <sup>2</sup> 3p <sup>1</sup>		<del></del>		
2. Write the full e	lectron confi	iguration (ex., 1s	2s1) of the follow	ing elements?
Nitrogen				
i. Silicon				
i. Calcium				
v. Strontium				
. Bismuth				
3. Write the abbro	eviated elect	ron configuratio	n (ex., [He] 2s <sup>1</sup> ) of	the following element?
Fluorine				
i. Magnesium			<del>-</del>	
ii. Selenium				
v. Rubidium			<del></del> -	
. Lead			_	
I. If each orbital of ollowing hold?	can hold a m	aximum of two	electrons, how mai	ny can each of the
. 2s ii	i. 5p	iii. 4f	iv. 3d	v. 4d
i. What is the sha	ape of an s o	rbital?		
5. What is the sha				

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**Robert R. Redfield** 

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