

Electron Configuration of Main Group Cations

Electron configuration

Na $1s^2 2s^2 2p^6 3s^1$

Orbital diagram

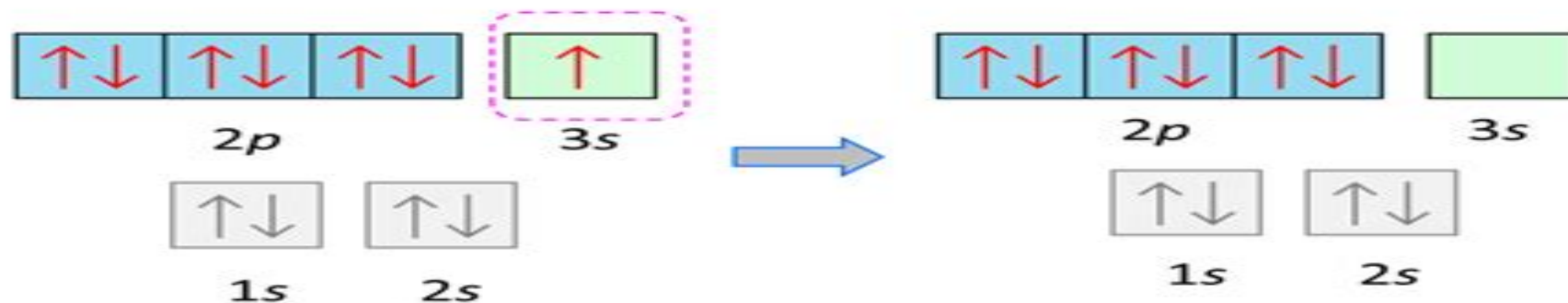


Focus on the valence shell

The charge is +1, therefore, one electron is lost:



The electron must be from the outermost valence orbital – $3s^1$



Na $1s^2 2s^2 2p^6 3s^1$

Na⁺ $1s^2 2s^2 2p^6$

↓ Same as

Ne $1s^2 2s^2 2p^6$



ChemistrySteps.com

Cations of main group metals attain the electron configuration of the noble gas that precedes them in the periodic table. They are **isoelectronic** with the previous noble gas.

Electron Configurations Of Atoms And Ions Lab Answers

EW Minium

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular gradient shape partially visible behind it.

Electron Configurations Of Atoms And Ions Lab Answers:

Whispering the Strategies of Language: An Mental Quest through **Electron Configurations Of Atoms And Ions Lab Answers**

In a digitally-driven world wherever monitors reign supreme and instant interaction drowns out the subtleties of language, the profound secrets and psychological subtleties hidden within words frequently get unheard. However, located within the pages of **Electron Configurations Of Atoms And Ions Lab Answers** a charming literary prize blinking with raw emotions, lies a fantastic quest waiting to be undertaken. Composed by a talented wordsmith, that marvelous opus attracts visitors on an introspective journey, delicately unraveling the veiled truths and profound influence resonating within ab muscles material of every word. Within the mental depths with this poignant review, we shall embark upon a heartfelt exploration of the book is primary themes, dissect its charming writing design, and fail to the effective resonance it evokes heavy within the recesses of readers hearts.

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