

Electron Configurations - KEY

Note: The electron configurations in this worksheet assume that lanthanum (La) is the first element in the 4f block and that actinium (Ac) is the first element in the 5f block. If your periodic table doesn't agree with this, your answers for elements near the f-orbitals may be slightly different (one more/less electron for the f-orbital).

- 1) sodium $1s^2 2s^2 2p^6 3s^1$
- 2) iron $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^6$
- 3) bromine $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^5$
- 4) barium $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2$
- 5) neptunium $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^6 7s^2 5f^5$
 also might be written (due to the close energy of d and f and similarities of elements versus predictable memorized config: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^6 7s^2 6d^1 5f^4$
- 6) cobalt $[Ar] 4s^2 3d^7$
- 7) silver $[Kr] 5s^2 4d^9$
- 8) tellurium $[Kr] 5s^2 4d^{10} 5p^4$
- 9) radium $[Rn] 7s^2$
- 10) lawrencium $[Rn] 7s^2 5f^{14} 6d^1$
 also might be written (due to the close energy of d and f and similarities of elements versus predictable memorized config: $[Rn] 7s^2 6d^1 5f^{14}$
- 11) $1s^2 2s^2 2p^6 3s^2 3p^4$ sulfur
- 12) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$ rubidium
- 13) $[Kr] 5s^2 4d^{10} 5p^3$ antimony
- 14) $[Xe] 6s^2 4f^{14} 5d^6$ osmium
- 15) $[Rn] 7s^2 5f^{11}$ einsteinium note: no d's, so the p. table referenced has Ac as 5f
- 16) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 4d^{10} 4p^5$ not valid (take a look at "4d"; must be 3 d not 4!)
- 17) $1s^2 2s^2 2p^6 3s^3 3d^5$ not valid (3p comes after 3s); also no 3 electrons in s
- 18) $[Ra] 7s^2 5f^8$ not valid (radium isn't a noble gas)
- 19) $[Kr] 5s^2 4d^{10} 5p^5$ valid
- 20) $[Xe]$ not valid (an element can't be its own electron configuration)

Electron Configuration Section 5 1 Answer Key

Atef Korchef



Electron Configuration Section 5 1 Answer Key:

, Chemistry John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant **Ebook: Chemistry: The Molecular Nature of Matter and Change** Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change **Super 10 Mock Tests for JEE Main 2018** Disha Experts, 2017-12-07 Super 10 Mock Tests for JEE Main contains 10 Mock Sample Tests designed exactly as per the latest pattern The book offers the BEST QUALITY Mock Tests with detailed solution to every question Answer keys and 100% solutions are provided along with cut off marks for each test The book also provides Trend Analysis of last 10 years Question Papers **Science (2023-24 KVS TGT)** YCT Expert Team , 2023 24 KVS TGT Science Solved Papers Practice Book **CRC Handbook of Chemistry and Physics** William M. Haynes, 2016-04-19 Mirroring the growth and direction of science for a century the Handbook now in its 93rd edition continues to be the most accessed and respected scientific reference in the world An authoritative resource consisting tables of data its usefulness spans every discipline This edition includes 17 new tables in the Analytical Chemistry section a major update of the CODATA Recommended Values of the Fundamental Physical Constants and updates to many other tables The book puts physical formulas and mathematical tables used in labs every day within easy reach The 93rd edition is the first edition to be available as an eBook *1998 Freshman Achievement Award* David R. Lide, 1998 *Objective Chemistry Vol 2 For Engineering Entrances 2022* , 1 Complete Study Pack for Engineering Entrances series provides Objective Study Guides 2 Objective Chemistry Volume 2 is prepared in accordance with NCERT Class 11th syllabus 3 Guide is divided into 25 chapter 4 complete text materials Practice Exercises and workbook exercises with each theory 5 Includes more than 5000 MCQs collection of Previous Years Solved Papers of JEE Main and Advanced BITSAT Kerala CEE KCET AP Objective Chemistry Volume 2 is divided into 25 chapters giving Complete Text Material along with Practice Exercises and Workbook exercises Chapter Theories are coupled with well illustrated examples helping students to learn the basics of Chemistry Housed with more than 5000 MCQs and brilliant collection of Previous Years Solved Papers of JEE Main and Advanced BITSAT Kerala CEE KCET AP TS EAMCET VIT and MHT CET which is the most defining part of this book Delivering the invaluable pool of study resources for different engineering exams at one place this is no doubt an excellent book to maximize your chances to get qualified at engineering entrances TOC Solid State Solutions Electrochemistry Chemical Kinetics Surface Chemistry Chemical Kinetics Surface Chemistry General Principle and Processes

of Isolation of Elements p Block Elements I Group 15 p Block Elements II Group 16 p Block Elements III Group 17 p Block Elements IV Group 18 d and f block Elements Coordinate Compounds Haloalkanes Haloarenes Alcohols Phenols Ether Aldehydes and Ketones Carboxylic Acids Amines Diazonium Salts Cyanides and Isocyanides Bimolecules Polymers Chemistry in Everyday Life Principles Related to Practical Chemistry JEE Advanced Solved Paper 2015 JEE Main Advanced Solved Papers 2016 JEE Main Advanced BITSAT Kerala CEE KCET AP TS EAMCET VIT MHT CET Solved Papers 2017 JEE Main Advanced BITSAT Kerala CEE KCET AP TS EAMCET VIT MHT CET Solved Papers 2018 JEE Main Advanced BITSAT Kerala CEE KCET AP TS EAMCET VIT MHT CET Solved Papers 2019 20

Handbook Of Porphyrin Science: With Applications To Chemistry, Physics, Materials Science, Engineering, Biology And Medicine (Volumes 6-10) Karl M Kadish, Roger Guilard, Kevin M Smith, 2010-06-29 This is the second set of Handbook of Porphyrin Science Porphyrins phthalocyanines and their numerous analogues and derivatives are materials of tremendous importance in chemistry materials science physics biology and medicine They are the red color in blood heme and the green in leaves chlorophyll they are also excellent ligands that can coordinate with almost every metal in the Periodic Table Grounded in natural systems porphyrins are incredibly versatile and can be modified in many ways each new modification yields derivatives demonstrating new chemistry physics and biology with a vast array of medicinal and technical applications As porphyrins are currently employed as platforms for study of theoretical principles and applications in a wide variety of fields the Handbook of Porphyrin Science represents a timely ongoing series dealing in detail with the synthesis chemistry physicochemical and medical properties and applications of polypyrrole macrocycles Professors Karl Kadish Kevin Smith and Roger Guilard are internationally recognized experts in the research field of porphyrins each having his own separate area of expertise in the field Between them they have published over 1500 peer reviewed papers and edited more than three dozen books on diverse topics of porphyrins and phthalocyanines In assembling the new volumes of this unique Handbook they have selected and attracted the very best scientists in each sub discipline as contributing authors This Handbook will prove to be a modern authoritative treatise on the subject as it is a collection of up to date works by world renowned experts in the field Complete with hundreds of figures tables and structural formulas and thousands of literature citations all researchers and graduate students in this field will find the Handbook of Porphyrin Science an essential major reference source for many years to come

CRC Handbook of Chemistry and Physics David R. Lide, 1995-03-09 This student edition features over 50 new or completely revised tables most of which are in the areas of fluid properties and properties of solids The book also features extensive references to other compilations and databases that contain additional information

Quantum Chemistry John P. Lowe, Kirk Peterson, 2011-08-30 Lowe's new edition assumes little mathematical or physical sophistication and emphasizes an understanding of the techniques and results of quantum chemistry It can serve as a primary text in quantum chemistry courses and enables students and researchers to comprehend the current literature This third edition has been thoroughly

updated and includes numerous new exercises to facilitate self study and solutions to selected exercises Assumes little initial mathematical or physical sophistication developing insights and abilities in the context of actual problems Provides thorough treatment of the simple systems basic to this subject Emphasizes UNDERSTANDING of the techniques and results of modern quantum chemistry Treats MO theory from simple Huckel through ab initio methods in current use Develops perturbation theory through the topics of orbital interaction as well as spectroscopic selection rules Presents group theory in a context of MO applications Includes qualitative MO theory of molecular structure Walsh rules Woodward Hoffmann rules frontier orbitals and organic reactions Develops MO theory of periodic systems with applications to organic polymers **Chemistry**

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Reactions of Metal Complexes in Solution Debabrata Banerjee, M K Bharty, 2022-09-21 Reaction Mechanisms of Metal Complexes in Solution provides a comprehensive overview of an often overlooked research area Despite its importance and recent reshaping of the field many inorganic chemists have lost an appreciation for the significance of stability constants and the thermodynamic aspects of complex formation Ideal for newcomers and established researchers in the field this book is a

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A-Level Chemistry E. N. Ramsden, 2000 This textbook has been updated to cover the new specifications for AS and A2 Chemistry and improved with new features and rewritten material to enhance learning and increase accessibility It covers all the main specifications for the English and Welsh Awarding Bodies and should be particularly suitable for students approaching A Level from GCSE Science Double Award This answer key is designed to support the core book and contains suggested answers worked solutions to the checkpoints and examination questions in the core book also synoptic questions for further practice complete with suggested answers and worked solutions to help develop confidence

Inorganic Chemistry in India, 2023-02-23 *Inorganic Chemistry in India* Volume 81 the latest release in the *Advances in Inorganic Chemistry* series highlights new advances in the field with this new volume presenting interesting chapters on topics such as Water Oxidation and Oxygen Reduction Reactions A Mechanistic Perspective Redox Activity as a Tool for Bond Activations and Functionalizations Heme Cu oxygen Intermediates of Amyloid beta Peptides associated with Alzheimer s Disease N Heterocyclic Silylene Coordinated Coinage Metal Complexes An Itinerary of Their Utilities Implications of strongly coupled catecholate based anchoring functionality of a sensitizer dye molecule towards photoinduced electron transfer dynamics and more Additional sections cover Application of Ru edta complexes in biomimetic activation of small molecules Kinetic and Mechanistic Impact and more Provides the authority and expertise of leading contributors from an international board of authors Presents the latest release in the *Advances in Inorganic Chemistry* series Updated release includes the latest

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web the c pole is a resonant antenna whereas the h pole is non resonant and must be matched with a tuner construction and materials used for the h pole are similar to the c pole i use non conductive glass fibre fishing rods beware of conductive

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