

Electron Configuration Practice
Chemistry

Name: _____

Due Date: _____

How to write an electron configuration:

- Determine the total number of electrons to be represented.
- Use the Aufbau principle to fill the orbitals with electrons for elements 1-23. Refer to electron configuration periodic table for elements after 23
- The sum of the superscripts should equal the total number of electrons. Example:
 $12\text{Mg } 1s^2 2s^2 2p^6 3s^2$

I. Configuration Writing Practice

- Write a ground state electron configuration for each neutral atom. Ground state means that all of the lowest possible energy levels (up to the proper number of electrons for the element) are filled.

1. Na: $1s^2 2s^2 2p^6 3s^1$

2. Pb: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^2$

3. Sr: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2$

4. U: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^6 7s^2 5f^3 6d^1$

5. N: $1s^2 2s^2 2p^3$

* 6. Ag: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1 4d^{10}$

7. Ti: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^2$

8. Ce: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^1 5d^1$

9. Cl: $1s^2 2s^2 2p^6 3s^2 3p^5$

10. Hg: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10}$

Electron Configuration Practice Packet

Kathy A. Zahler



Electron Configuration Practice Packet:

CliffsNotes Chemistry Practice Pack Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level RRB JE Navigator (PYQ & Practice Questions) CBT 2 (Electronics Engineering) Umesh Dhande, 2024-10-09 This comprehensive guide is designed to cater to the growing demand for accurate and concise solutions to RRB JE This book contains 2253 fully solved questions Including 2 PYQ RRB CBT 2 of Electronics Engineering 2 Shifts from RRB 2019 The book s key features include 1 Step by Step Solutions Detailed easy to follow solutions to all questions 2 Chapter Wise and Year Wise Analysis In depth analysis of questions organized by chapter and year 3 Detailed Explanations Clear explanations of each question ensuring a thorough understanding of the concepts 4 Simple and Easy to Understand Language Solutions are presented in a straightforward and accessible manner **Principles and Practice of Radiation Therapy: Physics, simulation, and treatment planning** Charles M. Washington, Dennis T. Leaver, 1996 Foundations of College Chemistry Morris Hein, Susan Arena, 2013-01-01 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years Foundations of College Chemistry Alternate 14th Edition has helped readers master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis Principles and Practice of Radiation Therapy Charles M. Washington, Dennis T. Leaver, 2004 The three separate volumes of the first edition each designed to stand alone have been combined into a single volume Several chapters have been consolidated and additional information added specifically in the ares of treatment planning electronic charting CT stimulation dose distribution and education Pedagogical features designed to enhance comprehension and critical thinking are incorporated into each chapter Elements include chapter outlines key terms and a glossary that includes significant terms from both editions Of particular note are the Review Questions and Questions to Ponder at the end of each chapter An Introductory Guide to EC Competition Law and Practice Valentine Korah, 1994 *Comprehensive Chemistry XI* Dr. B.

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