ELECTROCHEMISTRY

Basic Terminology, Construction of Electrochemical Cell and Its Repsentation

1	Which is not true for a standard hydrogen electrode?

(1) The hydrogen ion concentration is 1 M	(2) Temperature must be 25°C
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(1) chemical reaction produces electrical energy	(2) electrical energy produces chemical
reaction	

(1)
$$Zn^{2*}(s) + 2e^{-} \longrightarrow Zn(s)$$
 (2) $Zn(g) \longrightarrow Zn^{2*}(g) + 2e^{-}$

(3)
$$Zn^{2*}(g) + 2e^- \longrightarrow Zn(s)$$
 (4) $Zn^{2*}(aq.) + 2e^- \longrightarrow Zn(s)$

Electrochemisty Practical Manual

Lei Shi

Electrochemisty Practical Manual:

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