

## Electrochemistry-Ch.9 – PRACTICE TEST/EXAM QUESTIONS

### Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- \_\_\_\_ 1. In the reaction  
 $2HNO_{3(aq)} + Cu_{(s)} + 2H^{+}_{(aq)} \rightarrow 2NO_{2(g)} + Cu^{2+}_{(aq)} + 2H_2O_{(l)}$   
Which of the following statements is correct?  
a.  $HNO_3$  is the reducing agent, and  $Cu$  is the oxidizing agent  
b.  $HNO_3$  is the reducing agent, and  $H^{+}$  is the oxidizing agent  
c.  $Cu$  is the reducing agent, and  $H^{+}$  is the oxidizing agent  
d.  $Cu$  is the reducing agent, and  $HNO_3$  is the oxidizing agent  
e.  $H^{+}$  is the reducing agent, and  $Cu$  is the oxidizing agent
- \_\_\_\_ 2. The oxidation number of chromium ( $Cr$ ) in the dichromate ion ( $Cr_2O_7^{2-}$ ) is  
a. +2  
b. +4  
c. +6  
d. +7  
e. -2
- \_\_\_\_ 3. The oxidation number of fluorine ( $F$ ) in  $AlF_3$  is  
a. -3  
b. -2  
c. -1  
d. 0  
e. +1
- \_\_\_\_ 4. In  $Fe_2(SO_4)_3$  the oxidation numbers of  $Fe$ ,  $S$ , and  $O$  respectively are  
a. +2, +3, -4  
b. +3, +6, -2  
c. +2, +4, -8  
d. +2, +4, -2  
e. +4, +5, -3
- \_\_\_\_ 5. Which substance in the following reaction has undergone oxidation?  
 $2MnO_4^{-}(aq) + 5C_2O_4^{2-}(aq) + 16H^{+}(aq) \rightarrow 2Mn^{2+}(aq) + 10CO_{2(g)} + 8H_2O_{(l)}$   
a.  $MnO_4^{-}$   
b.  $C_2O_4^{2-}$   
c.  $H^{+}$   
d.  $Mn^{2+}$   
e.  $CO_2$
- \_\_\_\_ 6. This question consists of an assertion (statement) in the left-hand column and a reason in the right-hand column.

#### Assertion

Chlorine will cause iodine to form when chlorine is bubbled through a solution of potassium iodide.

#### Reason

Chlorine is a stronger oxidizing agent than iodine.

Select one of the following.

- if both assertion and reason are true statements, and the reason is a correct explanation of the assertion
- if both assertion and reason are true statements, and the reason is not a correct explanation of the assertion
- if the assertion is true but the reason is false
- if the assertion is false but the reason is a true statement
- if both assertion and reason are false statements

# Electrochemistry 2chapter Test A Answer Key

**Michael Seilmaier**



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Problems in Physical Chemistry JEE Main and Advanced Volume 2 Dr. RK Gupta, 2021-10-17

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