AP Chemistry Unit 3 Review

Lewis Diagrams, IMFs, and Gases

Honors Lecture Notes: Unit 2, Unit 3 Lectures 4-7

Lewis Diagrams

Steps to creating a Lewis Diagram:

1) Count the number of valence electrons

Place the least electronegative element in the center and single bond all terminal atoms to it

3) Complete all octets around the terminal atoms

4) Add any extra valence electrons to the central atom*

 If all valence electrons are used and the central atom is not full, create double or triple bonds

* Only atoms in period 3 or below can expand their octets

A resonance structure exists when there are multiple potential locations for a double bond. You draw structures with the double bond in each possible location, but in reality, the bonds are all identical and somewhere between single and double bonds in length and strength.

Formal Charge

To calculate the formal charge on an atom in a Lewis diagram, subtract the number of assigned electrons from the number of valence electrons for that atom. When counting assigned electrons, lone pairs count as two and bonds count as one.

Formal charges of greater than +1 or -1 do not exist in compounds. The less formal charge, the more stable (and thus, more likely) the compound is to exist. The total formal charge on all atoms in a polyatomic ion is equal to the overall charge on the ion.

VSEPR

Valence Shell Electron Pair Repulsion theory governs the shapes of molecules. A charge cloud is either a lone pair or any type of bond. Single/double/triple bonds all count as exactly one charge cloud. An outline of the shapes is below:

	0 Ione pairs	1 lone pair	2 Ione pairs	3 lone pairs
2 charge clouds	Linear			
3 charge clouds	Trigonal Planar	Bent		
4 charge clouds	Tetrahedral	Trigonal Pyramidal	Bent	
5 charge clouds	Trigonal Bipyramidal	See-Saw	T-Shaped	Linear
6 charge clouds	Octahedral	Square Pyramidal	Square Planar	

Shapes in the grey boxes are three-dimensional in nature.

Polarity.

Molecules are polar if they are asymmetrical. This is usually the case when the central atom has lone pairs on it (exception: square planar and linear molecules are symmetrical). The most electronegative element (usually the terminal atom, except when H is terminal) gets a negative dipole, and the least electronegative element gets a positive dipole. If the central atom has no lone pairs, the molecule can still be polar if the terminal atoms are different.

Hybridization

Single bonds are made up of sigma bonds. Double bonds are one sigma and one pi bond, and triple bonds are one sigma and two pi bonds.

Central atoms with 2 charge clouds exhibit sp hybridization. Central atoms with 3 charge clouds exhibit sp³ hybridization. Central atoms with 4 charge clouds exhibit sp³ hybridization.

Modeling Chemistry Unit 3 Review Answers

M Carnoy

Modeling Chemistry Unit 3 Review Answers:

<u>Introduction to Molecular Biology, Genomics and Proteomics for Biomedical Engineers</u> Robert B. Northrop,Anne N. Connor,2008-10-28 Illustrates the Complex Biochemical Relations that Permit Life to ExistIt can be argued that the dawn of the 21st century has emerged as the age focused on molecular biology which includes all the regulatory mechanisms that make cellular biochemical reaction pathways stable and life possible For biomedical engineers this concept is essential to

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centuries in Tibet accessible to help us all on our shared journey towards sustainable well being altruism inspiration and Lees' Loss Prevention in the Process Industries Frank Lees, 2012-11-05 Safety in the process industries is happiness critical for those who work with chemicals and hazardous substances or processes The field of loss prevention is and continues to be of supreme importance to countless companies municipalities and governments around the world and Lees is a detailed reference to defending against hazards Recognized as the standard work for chemical and process engineering safety professionals it provides the most complete collection of information on the theory practice design elements equipment regulations and laws covering the field of process safety An entire library of alternative books and cross referencing systems would be needed to replace or improve upon it but everything of importance to safety professionals engineers and managers can be found in this all encompassing three volume reference instead The process safety encyclopedia trusted worldwide for over 30 years Now available in print and online to aid searchability and portability Over 3 600 print pages cover the full scope of process safety and loss prevention compiling theory practice standards legislation case studies and lessons learned in one resource as opposed to multiple sources Recent Advances in Thermochemical Conversion of Biomass Ashok Pandey, Thallada Bhaskar, Michael Stöcker, Rajeev Sukumaran, 2015-01-28 This book provides general information and data on one of the most promising renewable energy sources biomass for its thermochemical conversion During the last few years there has been increasing focus on developing the processes and technologies for the conversion of biomass to liquid and gaseous fuels and chemicals in particular to develop low cost technologies This book provides date based scientific information on the most advanced and innovative processing of biomass as well as the process development elements on thermochemical processing of biomass for the production of biofuels and bio products on biomass based biorefinery The conversion of biomass to biofuels and other value added products on the principle biorefinery offers potential from technological perspectives as alternate energy The book covers intensive R D and technological developments done during the last few years in the area of renewable energy utilizing biomass as feedstock and will be highly beneficial for the researchers scientists and engineers working in the area of biomass biofuels biorefinery Provides the most advanced and innovative thermochemical conversion technology for biomass Provides information on large scales such as thermochemical biorefinery Useful for researchers intending to study scale up Serves as both a textbook for graduate students and a reference book for researchers Provides information on integration of process and technology on thermochemical conversion of biomass Comprehensive Medicinal Chemistry III, 2017-06-03 Comprehensive Medicinal Chemistry III Eight Volume Set provides a contemporary and forward looking critical analysis and summary of recent developments emerging trends and recently identified new areas where medicinal chemistry is having an impact The discipline of medicinal chemistry continues to evolve as it adapts to new opportunities and strives to solve new challenges These include drug targeting biomolecular therapeutics development of chemical biology tools data collection and analysis in silico models as predictors for biological

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