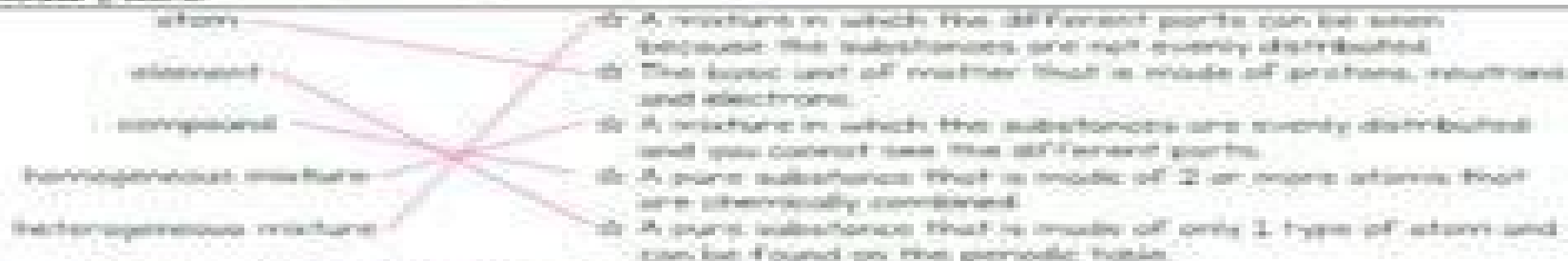


# PROPERTIES OF MATTER STUDY GUIDE



## ATOMS, ELEMENTS, COMPOUNDS + MIXTURES

Complete the vocabulary matching below by drawing a line to match the correct definition to the correct vocabulary word.



Complete the chart below about the subatomic particles and their charges:

	PARTICLE #1	PARTICLE #2	PARTICLE #3
NAME	proton	neutron	electron
CHARGE	+ (positive)	0 (neutral)	- (negative)
LOCATION	part of nucleus	part of nucleus	outer "cloud-like" orbit

Complete the graphic organizer below about matter below. Need help? Use the word bank to assist you!

heterogeneous	element	mixture	compound
	pure substance	electron	homogeneous



Classify the following examples as elements, compounds or mixtures in the table below.

carbon (C)	salt	water (H <sub>2</sub> O)
hydrogen (H)	salt water	sugar (sucrose)



ELEMENT	COMPOUND	MIXTURE
carbon (C) hydrogen (H)	water (H <sub>2</sub> O) salt (NaCl) sugar (sucrose)	salt

Why is carbon considered an element and carbon dioxide considered a compound?

Carbon is an element as the periodic table made of only carbon atoms. Carbon dioxide is carbon and oxygen chemically combined together.

# Matter Review Guide Answer Key

**M Mosston**



## **Matter Review Guide Answer Key:**

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