24. Here is a list of the types of bonds and interactions discussed in this section. Place them in order from the strongest to the weakest: hydrogen bonds, van der Waals interactions, covalent bonds, ionic bonds.



25. Use morphine and endorphins as examples to explain why molecular shape is crucial in biology.

#### Concept 2.4 Chemical reactions make and break chemical bonds

- Write the chemical shorthand equation for photosynthesis. Label the reactants and the products.
- For the equation you just wrote, how many molecules of carbon dioxide are there?

  How many molecules of glucose?

  How many elements in glucose?
- 28. What is meant by dynamic equilibrium? Does this imply equal concentrations of each reactant and product?

Testing Your Knowledge: Self-Quiz Answers

Now you should be ready to test your knowledge. Place your answers here:

1.\_\_\_\_2.\_\_\_3.\_\_\_4.\_\_\_5.\_\_\_6.\_\_\_7.\_\_\_8.\_\_\_\_

# **Holtzclaw Ap Biology Guide Answers Ch 16**

**David Baud** 

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