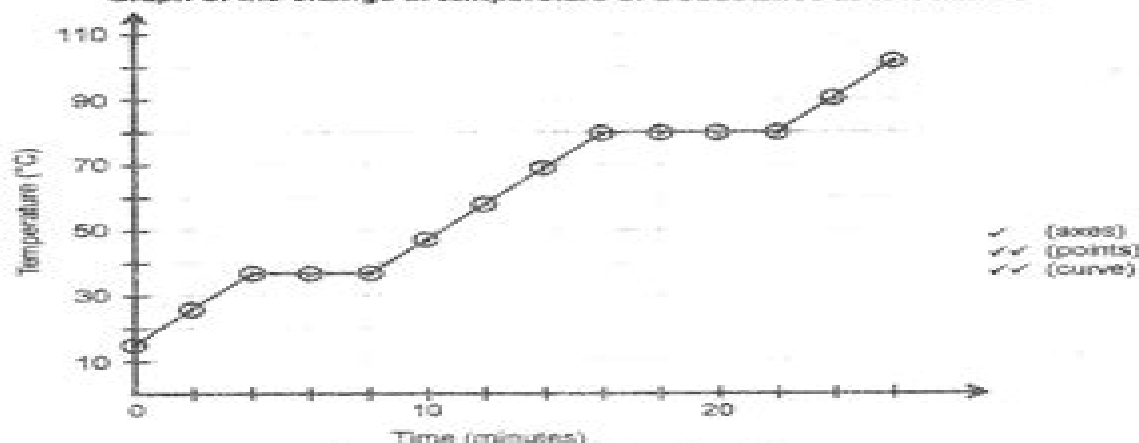


Graph of the change in temperature of a substance as it is heated

(5)



- 3.5 Substance exists as both a solid and a liquid ✓✓ or... is melting (2)
 3.6 37 °C ✓ (2)
 3.7 80 °C ✓ (1)
 3.8 An indication of the average kinetic energy ✓ of the particles in a substance. ✓ (2)
 3.9 Physical ✓ (1)
 3.10 In the solid state the intermolecular forces are the strongest and the particles cannot move. The heater provides thermal energy to the system, which is translated into kinetic energy (increase in temperature) until the intermolecular forces between the particles can be broken. ✓✓ At this point the thermal energy is transformed into latent energy to break the bonds, so no increase in temperature occurs. ✓✓ (4)

[23]

QUESTION 4

- 4.1
 4.1.1 Homogeneous ✓ (1)
 4.1.2 Homogeneous ✓ (1)
 4.1.3 Pure substance ✓ (1)
 4.2
 4.2.1 Potassium bromide ✓ (2)
 4.2.2 Calcium carbonate ✓ (2)
 4.2.3 Ammonium chloride ✓ (2)
 4.3
 4.3.1 ZnSO_4 ✓✓ (2)
 4.3.2 KMnO_4 ✓✓ (2)
 4.3.3 H_2O ✓✓ (2)

[17]

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1. Understanding the eBook Grade 11 Life Science Question Paper Of March 2014
 - The Rise of Digital Reading Grade 11 Life Science Question Paper Of March 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 11 Life Science Question Paper Of March 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 11 Life Science Question Paper Of March 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 11 Life Science Question Paper Of March 2014
 - Personalized Recommendations
 - Grade 11 Life Science Question Paper Of March 2014 User Reviews and Ratings
 - Grade 11 Life Science Question Paper Of March 2014 and Bestseller Lists
5. Accessing Grade 11 Life Science Question Paper Of March 2014 Free and Paid eBooks
 - Grade 11 Life Science Question Paper Of March 2014 Public Domain eBooks
 - Grade 11 Life Science Question Paper Of March 2014 eBook Subscription Services
 - Grade 11 Life Science Question Paper Of March 2014 Budget-Friendly Options
6. Navigating Grade 11 Life Science Question Paper Of March 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 11 Life Science Question Paper Of March 2014 Compatibility with Devices
 - Grade 11 Life Science Question Paper Of March 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 11 Life Science Question Paper Of March 2014
 - Highlighting and Note-Taking Grade 11 Life Science Question Paper Of March 2014
 - Interactive Elements Grade 11 Life Science Question Paper Of March 2014
8. Staying Engaged with Grade 11 Life Science Question Paper Of March 2014

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Grade 11 Life Science Question Paper Of March 2014
- 9. Balancing eBooks and Physical Books Grade 11 Life Science Question Paper Of March 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 11 Life Science Question Paper Of March 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Grade 11 Life Science Question Paper Of March 2014
 - Setting Reading Goals Grade 11 Life Science Question Paper Of March 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Grade 11 Life Science Question Paper Of March 2014
 - Fact-Checking eBook Content of Grade 11 Life Science Question Paper Of March 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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