

CHEMISTRY FINAL EXAM REVIEW GUIDE 2012 Scott

DISCLAIMER: THIS IS A SAMPLE OF QUESTIONS/PROBLEMS FROM THIS YEAR. PLEASE BE SURE TO REVIEW YOUR NOTES, LABS, COJ'S, OBJECTIVE SHEETS, ETC. DO NOT RELY ONLY ON THIS REVIEW GUIDE.

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1) Which of the following observations is quantitative?
a. The liquid turns blue litmus paper red.
b. The liquid boils at 100°C.
c. The liquid tastes bitter.
d. The liquid is cloudy.
- _____ 2) The symbols for units of length in order from largest to smallest are
a. m, cm, mm, km. c. km, mm, cm, m.
b. mm, m, cm, km. d. km, m, cm, mm.
- _____ 3) A volume of 1 milliliter is equivalent to
a. 1 cubic centimeter. c. 1 liter.
b. 1 gram. d. 10^{-3} cubic decimeters.
- _____ 4) Which pair of quantities determines the density of a material?
a. mass and weight c. volume and concentration
b. volume and weight d. volume and mass
- _____ 5) A sample of gold has a mass of 96.5 g and a volume of 5.00 cm³. The density of gold is
a. 0.0518 g/cm³. c. 101.5 g/cm³.
b. 19.3 g/cm³. d. 483 g/cm³.
- _____ 6) The number of grams equal to 0.5 kg is
a. 0.0005. c. 500.
b. 0.005. d. 5000.
- _____ 7) If 1 inch equals 2.54 cm, how many centimeters equal 1 yard?
a. 0.0706 cm c. 30.5 cm
b. 14.2 cm d. 91.4 cm
- _____ 8) A chemist obtained the values 5.2246 g, 5.2353 g, and 5.2501 g for the mass of a sample. Without knowing the true mass of the sample, it can be said that these values have
a. good precision.
b. good accuracy.
c. poor precision.
d. poor accuracy.
- _____ 9) The measurement 0.035550 g rounded off to two significant figures would be
a. 0.03 g. c. 0.036 g.
b. 0.35 g. d. 3.5×10^2 g.
- _____ 10) When 64.4 is divided by 2.00, the correct number of significant figures in the result is

Final Test Review Packet Chemistry

Thomas Griffiths



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Table of Contents Final Test Review Packet Chemistry

1. Understanding the eBook Final Test Review Packet Chemistry
 - The Rise of Digital Reading Final Test Review Packet Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Final Test Review Packet Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Final Test Review Packet Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Final Test Review Packet Chemistry
 - Personalized Recommendations
 - Final Test Review Packet Chemistry User Reviews and Ratings
 - Final Test Review Packet Chemistry and Bestseller Lists

5. Accessing Final Test Review Packet Chemistry Free and Paid eBooks
 - Final Test Review Packet Chemistry Public Domain eBooks
 - Final Test Review Packet Chemistry eBook Subscription Services
 - Final Test Review Packet Chemistry Budget-Friendly Options
6. Navigating Final Test Review Packet Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Final Test Review Packet Chemistry Compatibility with Devices
 - Final Test Review Packet Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Final Test Review Packet Chemistry
 - Highlighting and Note-Taking Final Test Review Packet Chemistry
 - Interactive Elements Final Test Review Packet Chemistry
8. Staying Engaged with Final Test Review Packet Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Final Test Review Packet Chemistry
9. Balancing eBooks and Physical Books Final Test Review Packet Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Final Test Review Packet Chemistry
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Final Test Review Packet Chemistry
 - Setting Reading Goals Final Test Review Packet Chemistry
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Final Test Review Packet Chemistry
 - Fact-Checking eBook Content of Final Test Review Packet Chemistry
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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