

# Answers to Topic questions

## Topic 1 Basic agricultural chemistry Answers

1

- a Molecules can be broken down chemically into the elements of which they are composed. Examples: carbon dioxide, water, sodium chloride, etc.
- b Isotopes atoms of the same element with different atomic weights. Examples: chlorine.
- c Valence electrons: outermost electrons around an atom.

2

The atom has a small, positively charged core called the nucleus. The atom is electrically neutral and consists of: i) electrons – negatively charged particles contained in a large outer volume called an electron cloud; ii) protons – positively charged particles concentrated in a small inner volume that contains most of the mass of the atom; iii) neutrons – electrically neutral and occur in the nucleus.

3

- a alkali-earth metal
- b fluorine / chlorine / bromine / iodine

4

- a covalent bonding
- b oxidation number
- c chemical formulae
- d gases
- e hydrogen bonding

5

- a
- b Hydrogen bonds that hold water molecules together are quite strong. But covalent bonds between atoms in a water molecule are stronger. Thus hydrogen bonds often break and reform which limits water's ability to flow.

6

- a carbon, oxygen
- b  $\text{z} \text{ off Fe, Mn, Ni, Mo, Cu, Zn, Cl, Cr}$

7

Plants use  $\text{CO}_2$  and  $\text{H}_2\text{O}$  to build complex organic compounds. They are enclosed by membranes during respiration.

8

- a alcohol
- b aliphatic
- c amide
- d sugar

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