

Activity B:**Beta decay**Get the Gizmo ready:

- Click **Reset**, and turn off **Show equation**.
- Select **Beta decay** from the **Type of decay** menu.
- Check that **Carbon** is selected.



Question: How does beta decay change the nucleus of a radioactive atom?

1. Observe: Click **Play** and watch the animation.

A. What happens to the decaying neutron during beta decay?

The neutron turns into a proton.

B. What is emitted from the nucleus during beta decay?

Gamma ray and electron

C. What is the mass number and charge of the emitted particle?

Mass number:

0

Charge:

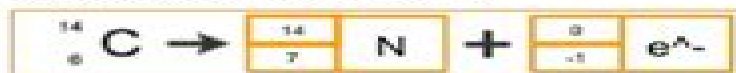
-1

2. Predict: During beta decay, a neutron is transformed into a proton and an electron (the **beta particle**), which is emitted. Gamma rays are often emitted during beta decay as well.

How will beta decay affect the atomic number and mass number of the atom?

The mass number will not change, but the atomic number will rise by one.

3. Calculate: Turn on **Write equation**. Fill in the first set of boxes with the mass number and atomic number of the daughter product and the next set of boxes with the mass number and atomic number of the beta particle. (Note: The atomic number of an electron is -1.)



Check your answer by turning on **Show equation** and clicking **Play**. Modify your equation if necessary. What isotope is produced by the beta decay of carbon-14? N-14

4. Practice: Turn off **Show equation**. Fill in the equations for the beta decay of iodine-131 and sodium-24 in the spaces below. Use the Gizmo to check your answers.



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Table of Contents Explorer Learning Nuclear Decay Key Answers

1. Understanding the eBook Explorer Learning Nuclear Decay Key Answers
 - The Rise of Digital Reading Explorer Learning Nuclear Decay Key Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Explorer Learning Nuclear Decay Key Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explorer Learning Nuclear Decay Key Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explorer Learning Nuclear Decay Key Answers
 - Personalized Recommendations
 - Explorer Learning Nuclear Decay Key Answers User Reviews and Ratings
 - Explorer Learning Nuclear Decay Key Answers and Bestseller Lists
5. Accessing Explorer Learning Nuclear Decay Key Answers Free and Paid eBooks
 - Explorer Learning Nuclear Decay Key Answers Public Domain eBooks
 - Explorer Learning Nuclear Decay Key Answers eBook Subscription Services
 - Explorer Learning Nuclear Decay Key Answers Budget-Friendly Options

6. Navigating Explorer Learning Nuclear Decay Key Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Explorer Learning Nuclear Decay Key Answers Compatibility with Devices
 - Explorer Learning Nuclear Decay Key Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explorer Learning Nuclear Decay Key Answers
 - Highlighting and Note-Taking Explorer Learning Nuclear Decay Key Answers
 - Interactive Elements Explorer Learning Nuclear Decay Key Answers
8. Staying Engaged with Explorer Learning Nuclear Decay Key Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explorer Learning Nuclear Decay Key Answers
9. Balancing eBooks and Physical Books Explorer Learning Nuclear Decay Key Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explorer Learning Nuclear Decay Key Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Explorer Learning Nuclear Decay Key Answers
 - Setting Reading Goals Explorer Learning Nuclear Decay Key Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Explorer Learning Nuclear Decay Key Answers
 - Fact-Checking eBook Content of Explorer Learning Nuclear Decay Key Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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