### Explorizinamina

Name:	Diate:

# Student Exploration: Nuclear Decay

Vocabulary: siphs particle, storric number, beta particle, daughter product, garness tay, isotope, mass number, nuclear decay, positror, radioactive, substorric particle.

Prior Knowledge Questions (Do these BEFCIPE using the Giams.)

The chart below gives the locations, charges, and approximate masses of three substantial particles. The approximate mass of each particle is given in universal mass units 0.6.

Particle	Location	Charge	Approximate mass
Protein.	Nucleus	S 755	1.0
Neutron	Nucleus	0	1.0
Destron	Orbitale	47.0	Du

The mass number of an atom is equal to the sum of protons and neutrons in the nucleus.

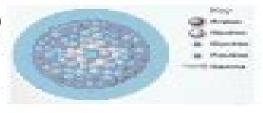
A helium atom has 2 protons and 2 neutrons. What is the mass number of this atom?

The attention surebox of an element is equal to the number of pretons in each atom of the element. All helium atoms have 2 protons. What is the atomic number of helium?

### Gilamo Warm-up

While rest atoms are stable, some are removetive, which means that they have a tendency to undergo spontaneous nucleus decay. The decay of radioactive atoms generally results in the emission of particles and/or energy.

Several types of nuclear decay can be explored with the Aluctear Decay Sizmo \*\*. On the Gizmo, shock that Alpha decay and Unonium are selected.



Click Play (CEX) and then click Pause (CEX) when the alpha particle is clearly visible.

What is an alpha particle made of?

2. Click Play and observe. Besides the alpha particle, what else is emitted from the nucleus.

during algha decay?

German rays are energetic electromagnetic waves; they are often emitted in nuclear decay.

# **Explore Learning Student Exploration Nuclear Decay Answers**

**Michael Rajnik** 

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