

ELEMENTS COMPOUNDS AND MIXTURES



Types: _____ Date: _____

Elements and Compounds

Mixtures vs. Compounds

Directions: Complete whether each mixture described is uniform or nonuniform. Write "U" for uniform and "N" for nonuniform.

1. Air	Answer: _____
2. Juice	Answer: _____
3. Table Salt (NaCl)	Answer: _____
4. Cement	Answer: _____
5. Lemonade	Answer: _____
6. Sugar	Answer: _____
7. Carbon Dioxide (CO ₂)	Answer: _____
8. Water	Answer: _____
9. Fruit Mix	Answer: _____
10. Butter	Answer: _____
11. _____	Answer: _____

Name: _____ Class: _____

Elements and Compounds

Instructions:
 1. If it is an element, write down its chemical symbol.
 2. If it is a compound, write down its chemical formula.

Ex. Oxygen

Type	Chemical Symbol
Element	O

1. Water

Type	Chemical Symbol

2. Carbon Dioxide

Type	Chemical Symbol

Name: _____

Elements and Compounds

Element

Directions:

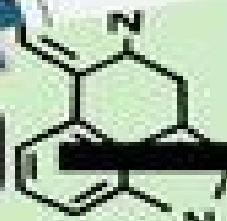
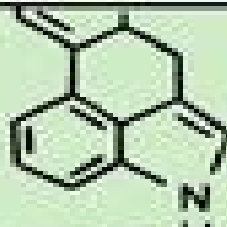
- Definition: A pure substance made up of only one type of atom.
- Examples: Hydrogen (H₂), Oxygen (O₂), Carbon (C), Sodium (Na), Iron (Fe).

Chemical Symbol:

- Definition: A shorthand representation of an element consisting of one or two letters.
- Examples: H (Hydrogen), O (Oxygen), Fe (Iron), Au (Gold).

Examples of Elements:

- Hydrogen (H₂)
- Oxygen (O₂)
- Carbon (C)
- Sodium (Na)
- Iron (Fe)
- Gold (Au)



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Roman Wölfel



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