

Activity A:

Subatomic particles

Get the Gizmo ready:

- Use the arrows to create an atom with two protons, two neutrons, and two electrons.
- Turn on **Show element name**.



Question: What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on **Show element symbol** and **Element notation**. Three numbers surround the element symbol: the **mass number** (A), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** (Z).
2. **Investigate:** Watch how the numbers change as you add or remove particles.



- A. Which number is equal to the number of protons in the atom?
- B. How can you calculate the number of neutrons (N) in an atom?
- C. Which particle (proton, neutron, or electron) has a
Positive charge?
Negative charge?

neutrons

by the number of protons

electron

proton

Electrical charge?

3. **Explore:** Use the Gizmo to create different atoms of the element Helium, which has an atomic number of 2. Each Helium atom has 2 protons. Change the number of neutrons and electrons. The number of neutrons is the mass number minus the atomic number. Each Helium atom has 2 protons, 2 neutrons, and 2 electrons. Helium is a noble gas, which means it does not react with other elements.

- A. What are the mass numbers of Helium?
- B. What are the mass numbers of Helium?
- C. What are the mass numbers of Helium?

2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14, 15, 16, 17, 18, 19, 20, 21, 22, 23, 24, 25, 26, 27, 28, 29, 30, 31, 32, 33, 34, 35, 36, 37, 38, 39, 40, 41, 42, 43, 44, 45, 46, 47, 48, 49, 50, 51, 52, 53, 54, 55, 56, 57, 58, 59, 60, 61, 62, 63, 64, 65, 66, 67, 68, 69, 70, 71, 72, 73, 74, 75, 76, 77, 78, 79, 80, 81, 82, 83, 84, 85, 86, 87, 88, 89, 90, 91, 92, 93, 94, 95, 96, 97, 98, 99, 100

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2

4. **Explore:** Use the Gizmo to answer the following questions.

- A. How many electrons are in a neutral atom of Helium?
- B. How many electrons are in a neutral atom of Helium?
- C. What is the mass number of an atom with 2 protons and 2 neutrons?
- D. What is the mass number of an atom with 2 protons and 2 neutrons?
- E. How many electrons are in Helium?

2

2

4

4

2

Element Builder Gizmo Exploration Sheet Answer Key

Karin Nielsen-Saines

A red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the author's name.

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