

## Electron Configuration Practice Worksheet **KEY**

*In the space below, write the unabbreviated electron configurations of the following elements:*

- 1) sodium       $1s^2 2s^2 2p^6 3s^1$
- 2) iron       $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^6$
- 3) bromine       $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^5$
- 4) barium       $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2$
- 5) neptunium       $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^6 7s^2 5f^4$

*In the space below, write the abbreviated electron configurations of the following elements:*

- 6) cobalt       $[Ar] 4s^2 3d^7$
- 7) silver       $[Kr] 5s^2 4d^9$
- 8) tellurium       $[Kr] 5s^2 4d^{10} 5p^4$
- 9) radium       $[Rn] 7s^2$
- 10) lawrencium       $[Rn] 7s^2 5f^{14} 6d^1$

Determine what elements are denoted by the following electron configurations:

- 11)  $1s^2 2s^2 2p^6 3s^2 3p^4$  **Sulfur**
- 12)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$  **Rubidium**
- 13)  $[Kr] 5s^2 4d^{10} 5p^3$  **Antimony**
- 14)  $[Xe] 6s^2 4f^{14} 5d^8$  **Osmium**
- 15)  $[Rn] 7s^2 5f^{11}$  **Einsteinium**

Determine which of the following electron configurations are not valid:

- 16)  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 4d^{10} 4p^5$       **should be  $3d^{10}$**
- 17)  $1s^2 2s^2 2p^6 3s^2 3d^5$        **$3s^2$  is impossible & "p" instead of "d"**
- 18)  $[Ra] 7s^2 5f^8$       **Ra is not a noble gas**
- 19)  $[Kr] 5s^2 4d^{10} 5p^5$       **Valid**
- 20)  $[Xe]$       **Element can't be its own electron configuration.**

# Electron Configuration Worksheet And More Answers

**Terry C. Jones**



## **Electron Configuration Worksheet And More Answers:**

Language Across the Curriculum Mr. Rohit Manglik, 2023-03-23 In this book we will study about the role of language in understanding subject content and promoting literacy across disciplines Class 10th Science Worksheet, This book is as per the guidelines syllabus and marking scheme issued by CBSE for Class X The salient features of this workbook are The questions in the this book have been so designed that complete syllabus is covered This book help students to identify their weak areas and improve them Additional it will help students gain confidence The questions in the book are of varying difficulty level and will help students evaluate their reasoning analysis and understanding of the subject matter **Educart**

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