

# Electrolysis, the Faraday, and Avogadro's Number

1. Final volume in buret 24
2. Initial volume in buret 44.0 ml
3. Volume of hydrogen 20
4. Temperature of solution 23°C
5. Height of water column 33 mm
6. Barometric pressure 740 mm Hg
7. Vapor pressure of  $H_2O$  (see Appendix "Answers to Selected Pre-lab Questions") 23.8 mm Hg
8. Pressure of  $H_2$  \_\_\_\_\_ mm Hg \_\_\_\_\_ atm (show calculations)

9. Moles of  $H_2$  produced (show calculations) \_\_\_\_\_

10. Time reaction started 0

~~11. Time reaction terminated~~ \_\_\_\_\_

12. Elapsed time 1 min 50 s

13. Current 1.2 A

14. Number of coulombs passed \_\_\_\_\_ C

15. Value for faraday \_\_\_\_\_ Accepted value \_\_\_\_\_ (show calculations)

# Electrolysis Experiment Lab Answer Key

**Ying-Ying Zheng**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Electrolysis Experiment Lab Answer Key:**

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