

Electrochemistry Note Taking Guide Answers

Frederick Remsen Hutton



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Spangler, Arthur Maurice Greene, Stewart McCulloch Marshall, 1906 **Introduction to the Rarer Elements** Philip
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web for example take the reaction of lead ii nitrate with hydrochloric acid to form lead ii chloride and nitric acid shown below

$$\text{pb no}_3 \text{ 2 aq } + 2 \text{ hcl aq } \rightarrow \text{pbcl}_2 \text{ s } + 2 \text{ hno}_3$$

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web you may have to consult the solubility rules write the net ionic equation for the reaction of $\text{bacl}_2 \text{ aq}$ and $\text{na}_2\text{so}_4 \text{ aq}$ you may have to consult the solubility rules write

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web choose the correct net ionic equation representing the reaction that occurs when solutions of potassium carbonate and copper i chloride are mixed a $\text{K}_2\text{CO}_3 \text{ aq} + 2\text{CuCl} \text{ aq}$

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