# FORMULAS WITH POLYATOMIC IONS

Name \_\_\_\_\_

Matching the horizontal and vertical axes, write the formulas of the compounds with the following combination of ions. The first one is done for you.

|      | OH-                       | NO <sub>3</sub> - | CO <sub>3</sub> -2             | SO <sub>4</sub> -2 | PO <sub>4</sub> -3 |
|------|---------------------------|-------------------|--------------------------------|--------------------|--------------------|
| н•   | HOH<br>(H <sub>2</sub> O) | HNO <sub>3</sub>  | H <sub>2</sub> CO <sub>3</sub> | H₂SO₄              | H₃PO₄              |
| Na⁺  |                           |                   |                                |                    |                    |
| Mg+2 |                           |                   |                                |                    |                    |
| NH₄⁺ |                           |                   |                                |                    |                    |
| Ca+2 |                           |                   |                                |                    |                    |
| К+   |                           |                   |                                | = -                |                    |
| Al+3 |                           |                   |                                |                    |                    |
| Pb+4 |                           |                   |                                |                    |                    |

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